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An Evaluation of
Future Coal Technology

by
Louis B. Mehl

October 1978

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AN EVALUATION OF FUTURE COAL TECHNOLOGY

BY

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THESIS

Submitted in partial fulfillment of the requirements
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1.0 INTRODUCTION

This research study was designed to be part of an input-output regional energy demand study of the Ohio River basin to the year 2000. In turn, the demand study is a supplementary part of the U.S. Environmental Protection Agency funded impact assessment of energy conversion facilities in the Ohio River Basin Energy Study (ORBES) region. The purpose of the present study is the establishment of the technical input-output coefficients for existing and probable future coal utilization technologies. These will be integrated into an I/O format to which subsequent alterations may be readily made. The present study entails a three-phase operation.

The first phase is the determination of those processes which can be commercially available by the year 2000; the basic constraint is the time horizon. For example, Figure 1.1 (1,p.935) demonstrates the normal procedure involved in bringing a coal gasification technique to commercial application. Some approximate relative costs are also shown. The time spans for plant testing have been shown, based on past performance, to be rather optimistic. They appear to represent a minimum time requirement. Therefore, the initial list of coal conversion processes includes only those that are already being tested at least on a pilot plant scale.

"Meeting the nation's energy needs in 2000 requires commercial demonstration by 1985 of a spectrum of first and second-generation technologies for the production of high and low Btu gas, distillate, residuals, solvent-refined coal, and methanol so that informed investment and environmental decisions for achieving the 2000 goal can be made'." (2,p.76).

This is illustrated in Table 1.1 (2,p.76)

The initial list of possible processes is rather large. However, all must be considered feasible by the year 2000. These processes are shown in Tables 1.2, 1.3, and 1.4. Much of the information for these tables was obtained from Sargent and Lundy (3). This was updated where needed.

The second phase is the detailed cost analysis of the various techniques. This involves not only projections of the original capital costs, but an analysis of the relative operating costs of a commercial-sized plant. The magnitude of the various problems, and their consequences, are illustrated for the various processes. The detailed analysis included in the first phase is augmented in the second. Finally, the most realistically available techniques are chosen based on the economics involved.

Tables 1.2, 1.3 and 1.4 deal only with coal gasification and liquefaction. Obviously there are other coal utilization

technologies: MHD, fuel cells, combined cycle generation, and direct pyrolysis. However, much of the work in these areas is dependent on prior improvements in coal gasification; improvements in conjunction, or competition, with coal liquefaction.

Tables 1.2, 1.3 and 1.4 demonstrate the fundamental breakdown of the processes considered; low and high Btu gasification, and coal liquefaction. Separating low and high Btu gasification essentially differentiates between the primary characteristics in gasification processes: type of reactor, coal handling, and scale of use. Coal liquefaction techniques are maintained under a single heading because the major difference in these processes is between indirect and direct hydrogenation with Fisher-Tropsch the only major process using indirect hydrogenation.

A synopsis of the processes listed in Tables 1.2-1.4 is given, graphically when available, as well as their potential products. But the primary purpose of the analysis is to set forth those considerations which influence the feasibility of their commercial development. In conjunction with this, two different ratings are given in the tables. The first is the Status Rating (SR). This is a relative listing of the availability of the processes according to their 1978 status. The general breakdown of the rating system is:

1. Commercially available and used (or under construction).
2. Commercially available but not used.
3. Demonstration plant testing
4. Demonstration plant construction.
5. Pilot plant testing

The second, more subjective, rating is the Feasibility Rating (FR), which compares the feasibility of the processes in relation to the primary advantages or the shortcomings involved in conversion. The guidelines for this analysis are:

1. Refinement of current process
2. Problems most likely to be resolved
3. Testing limitations of process
4. Process used regardless of shortcomings
5. Problems inherent to process and not used at present

Not all of the specific processes listed are discussed here because some of them are merely extensions or combinations of other methods. Furthermore, those processes for which there was

insufficient data for comment were simply listed. In many cases, where data are unavailable, the omission is not due to proprietary disclosure but, rather, to the fact that the process is, at best, in a very formative stage.

It is rather obvious that the above discussion is preoccupied with coal conversion technologies. This will also later be evidenced by the separate comment section for the coal conversion chapter. However, this is because many of the technological and economical considerations involved in coal conversion reflect directly on other coal utilization technologies. For example, the aforementioned MHD process with coal is dependent on an efficient gasification process. This dependency is especially true of the high-sulfur coal in this geographic region of study. Consider direct combustion of coal in a fluidized medium - this could be an efficient way to handle the caking coals of this region. In addition, economic ways of handling the sulfur products could make combustion environmentally desirable. But hot-gas cleanup of coal gasification products would make a combined cycle system attractive. There is then the relative costs of transporting this power to and/or throughout the region. These alternatives are all considered in subsequent chapters, but the future of coal utilization - specifically high-sulfur, caking coal in the ORBES region - is considered with coal conversion technology as a baseline.

TABLE 1.1 Plants Required for Gas-from-Coal Industry					
By 1985	Unit Plant Output	No. Plants	Unit Cost (1977 dollars)	Capital Investment \$ billion (1977 dollars)	
Pioneer and demonstration plants					
High-BTU pipeline gas	250 Mbt/d	3	1.8 billion		4.2
Lurgi technology	80 Mbt/d	3	400 million		1.2
Advanced second-generation technology	50 billion BTU/day	3	250 million		0.75
Low BTU gas					
Total					6.15
By 2000					
High-BTU gas plants	250 Mbt/d	60	1.1 billion		66
Low-BTU gas plants	150 billion BTU/day	30	600 million		18
Total					84

Source: Gas Research Institute (2.p. 76)

PROCESS	OFFERED	NAME	BT	PROCESS	STATUS	COAL	PROJECT	BTU	EFF.	\$/MBTU	O ₂	NEEDED	REACTOR	PHASE	SN	PN
KOPPELS- TOLZEL	---	---	---	Gasification	Available	Commercially	---	302	77	---	Yes	Atmos.	2	5		
LUKCI	LUKCI	LUKCI	La. Nat. (cont.)	Fixed Bed	Commercially	Sub-Bil.	250	302	78	---	Yes	500	1	4		
UCAS	UCI	Chicago	---	Fluidized	Pilot	Caking	840	136	71	---	No	---	5	2		
WILLMAN- GALUSHA	McKOWELL	York, Pa.	---	Fixed Bed	Commercially	---	150	160	80	---	---	Atmos.	2	4		
WATLER	DAY POWER	---	---	Atmospheric	Commercially	All U.S.	8	TPH	Low	91	---	Atmos.	4	3		
ADVANCED	WECO	Waltz Mill, Penn.	---	Fluidized	Proposed	Caking	15	130	---	---	No	15	5	3		
GOULFICA- TILION	---	---	---	Fluidized	Design	Scale-up in	---	---	---	---	---	---	---	---		
SYSTEM FOR REC- TICAL	---	---	---	Bed Gasifier	Design	Scale-up in	---	---	---	---	---	---	---	---		
CONSHA- TILION	---	---	---	Bed Gasifier	Design	Scale-up in	---	---	---	---	---	---	---	---		

TABLE 1.2
Low-Btu Coal Gasification Status

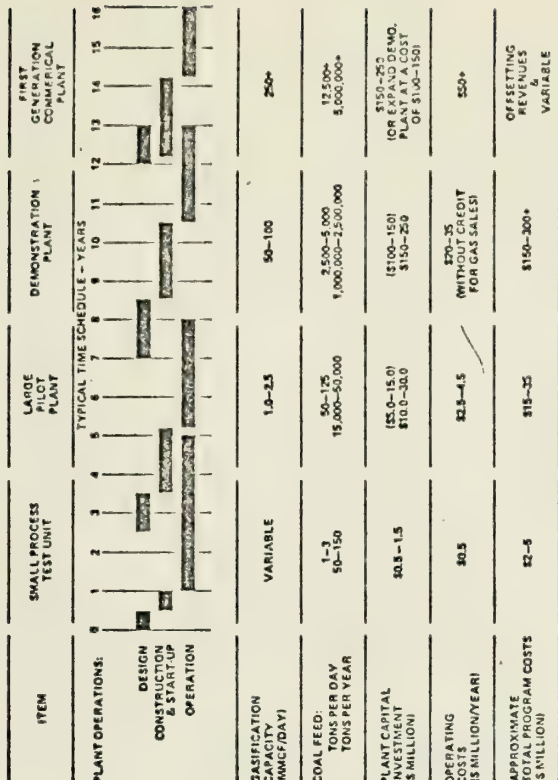
PROCESS	OFFERED	NAME	BT	PROCESS	STATUS	COAL	PROJECT	BTU	EFF.	\$/MBTU	O ₂	NEEDED	REACTOR	PHASE	SN	PN
BI-CAS	Bituminous	Homestead, Pa.	---	Coal	Pilot	All U.S.	120	943	69	3.50	Yes	1000	5	2		
CO ₂ ACCEPTION	Dev. Co.	So. Dakota	---	Coal	Pilot	Sub-Bil.	40	953	63	3.50	No	150-300	5	3		
LOCAS	LOCU	Ill.	---	Coal	Demo.	All U.S.	2000	950	65	---	No	50	4	2		
UCAS	UCI	Chicago	---	Fluidized	Operating	All U.S.	75	940	71	2.75	Yes	1000	5	2		
STAGGINS	Com. Coal	Ohio	---	Modified	Demo.	All U.S.	3387	950	63	---	---	---	4	4		
STIMULANT	Lummas Co.	Brinceton, Pa.	---	Two Zone Fluid-	Pilot	All U.S.	72	927	65	4.50	Yes	1000	5	2		

TABLE 1.3
High-Btu Coal Gasification Status

Steam-2
Steam-1 from
Electrothermal and Steam-1 from

TABLE 1.4 Coal Liquefaction Status									
PROCESS NAME	BY	STATUS	PROCESS TYPE	PRODUCT	PRICE	PH	SM	1	5
LEUDB	PMC	M.J., Princeton	Hydrogenation of COGAS, Pilot plant, Feedstock derived from coal	Pyrolysis	36 TPD	---	---	4	2
CLEAN COKE	U.S. Steel	PA, Monroeville	Fluidized bed, Feedstock derived from coal	Pyrolysis	1/2 TPD	---	---	5	2
EDMUND	Exxon	Baytown, TX	Solvent extraction, Feedstock derived from coal	Solvent extraction	250 TPD	55-75	---	5	3
THOMPSON	U.S. Africa	Commercially available	Solvent extraction, Feedstock derived from coal	Synthesis	---	55-75	---	1	5
TRICOL	Research Corp., Calif.	Similar to SRC, Feedstock derived from coal	Solvent extraction, Feedstock derived from coal	Synthesis	600 TPD	60-70	---	4	2
SHC	Energy and Minerals, West Va.; Mineral Co., Morgantown, Ky.	Similar to SRC, Feedstock derived from coal	Solvent extraction, Feedstock derived from coal	Synthesis	600 TPD	60-70	---	4	2

FIGURE 1.1



Timetable for R & D Work to Culminate in Commercial Applicability of Coal Gasification Facility

Source: (1,p.935)

2.0 COAL CONVERSION

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- 1. "Coal Technology." Modern Energy Technology. Vol. 2. Research and Education Association 1975.
- 2. Linden, H.R., "All-out push urged for U.S. gas supply " The Oil and Gas Journal November 21, 1977.

2.1 Coal Gasification

In general, coal gasification processes involve the following steps:

- Transport
- Preparation
- Preheat
- Gasification
- Tar Removal
- Shift Conversion
- Purification
- Methanation
- Transport and Storage

The principal differences among coal gasification processes are the manner in which coal is admitted to the gasifier, the type of reactor bed, and the gasification heat source. These differences may be dictated by the type of coal available. A generalized flow diagram with accompanying reactions is shown in Figure 2.1 (5.p.45) for the gasification process. If air is used in the hydrogasification step, a low-Btu gas is the result; if

oxygen is used, an intermediate-Btu gas is the product. If a high-Btu gas is desired, shift conversion and methanation steps are required. One of the desirable features of coal gasification is the ability to purify the gas with proven technology by removing the H_2S . The main disadvantage of this feature, however, is that currently the gas must first be cooled, which lowers the overall efficiency of the process.

Figure 2.2 (6.p.27) locates major water and coal resource regions, while Figure 2.3 (6.p.33) indicates potential areas for coal gasification. The ORBES region has abundant high-sulfur coal reserves for both deep and strip mines. Overall there is a large developed mining industry and a considerable market for a coal gasification product. However, the economic factors are at least as important as raw-material existence. Large blocks of reserves, such as bituminous coal in Illinois, would help keep costs down for such a capital-intensive process [39].

2.1.1 Low and Medium-Btu Coal Gasification

"There are three main reasons for developing low-Btu gas technology: to develop a gasifier that can economically handle large volumes of gas throughput; to achieve high conversion of coal to gas without any liquid by-products; and to develop a process that will gasify coal with a minimum of damage to the environment." (7.p.51)

Low and medium-Btu gasification processes are seen primarily as an inexpensive and clean utility boiler fuel. The processes have some inherent advantages [49]:

1. They can produce gas using state-of-the-art technology.
2. They are less complicated and hence, less expensive than high-Btu processes.
3. They are amenable to modular expansion and ease in sizing.
4. The gas cleanup can be designed directly into the system and meet present standards.

Electric utilities, large-scale plants, and industrial parks, are feasible applications for a low-Btu gasification product. For example, there are no major technological constraints to the use of a medium-Btu gas in a secondary steel plant or in using a low-Btu gas in a refinery [30]. A study by Gilbert Associates (8.p.4) has shown that low-Btu gasification is the most economical coal conversion process for industries requiring up to 50 million Btu/day. For uses requiring more than this, a fixed bed pressurized system producing medium-Btu gas becomes lower in cost. At 150 billion Btu/day, an atmospheric entrained bed gasifier producing medium-Btu gas appears to be the most economical. For an output of 100 billion Btu/day, they found that a low-Btu fixed bed gasifier has the lowest initial cost, with the operating costs for fixed-bed, fluidized-bed, and

entrained-bed systems being about the same. Four of the feasible techniques listed in Table 1.2 are commercially available. Three of them are of the fixed bed variety. This is indicative of the initial emphasis placed on utilizing second generation techniques for producing a "pipeline-quality" gas.

2.1.1.1 Lurgi (9,10,11,28,35,48)

This technique is a fixed bed, pressurized, gasifier developed in Germany some 40 years ago. It is the archetypical first generation technique. Figure 2.4 (11,p.80) illustrates the coal path. High-pressure operation enhances methanation.

Scaling up is a major obstacle for this process. Additionally, the fixed bed process has trouble with eastern caking coals and coals containing fines. The temperature must be strictly controlled so that the ash fusion temperature is not reached. This limits process efficiency and explains why current plans for construction are all done outside the ORBES region where non-caking coals are abundant (28,46). However, worldwide, many different coals have been used with this process. In addition to the above, leading to high initial investment, operating costs are increased by the required use of oxygen. However, air could be used to produce a raw gas with the

composition shown in Table 2.1 (11,p.80).

2.1.1.2 Wellman-Galusha (11,12,48)

This is another fixed bed reactor process that operates at atmospheric pressure with an agitator to help maintain a uniform bed. Its raw gas composition is shown in Table 2.2 (11,p.87).

All types of coals can be used if the agitator is applied but, again, this system must be operated below the ash fusion temperature, reducing efficiency. Its low operating pressure limits its possible utilization. Although this gasifier has been operated commercially for many years, its limited reactor size deters future applications. Tests have been run on a pressurized Wellman-Galusha gasifier, but efficiency is still limited by fusion temperature.

2.1.1.3 Koppers-TOTZEX (6,9,10,11,48)

This first generation, proven process, was developed in 1948. Although no plants exist in this country, this suspension-bed process seems to have some advantages over Lurgi.

Coal size is not a limiting factor, and caking coals can be handled without pretreatment. Figure 2.5 (11,p.81) illustrates the entire process. Figure 2.6 illustrates a simplified design of the reactor. The slagging system increases throughput over a fixed-bed system. In addition slagging temperatures increase steam conversion and pressurization enhances methanation.

Fly-ash removal has been a problem with this technique. The major drawback is that it requires almost 30 percent more oxygen than the Lurgi process for an equivalent amount of $\text{CO} + \text{H}_2$. Air cannot be used because temperatures would not be high enough for a slagging operation. Again scale-up could be a problem. Typical raw gas compositions are shown in Table 2.3 (11,p.81).

2.1.1.4 WINKLER (10,11,17,48)

The Winkler process is the only commercially-proven fluidized bed process. It was first proposed in 1922. Its advantage is that this process can be used on caking and high ash coal. Furthermore, no tars or oils are produced. The process is relatively easy to operate. However, it is an atmospheric process, and entrained coal must be recycled for good efficiency. As in all fluidizing processes, coal reactivity is a strong determinant in process efficiency. Partial oxidation may be

required for caking coals. A flow diagram of the process, and its raw gas composition, are given in Figure 2.7 and Table 2.4, respectively (11,p.86).

2.1.1.5 U-Gas (3,7,11)

This is one of the few second generation fluidized bed techniques considered feasible for low-Btu gasification by the year 2000. The process flow is shown in Figure 2.8 (7,p.51).

This technique has the desirable low-Btu process features of not producing tars and oils, having a simple design, and the ability to handle all types of coal. However, pretreatment is required for caking coals. A drawback of many fluidized systems is their inability to handle low carbon-content ash. This is essential in order to maintain non-slugging operation. Thus, pilot work is being done whereby the ash-agglomerating technique (AAG) is incorporated into the gasifying region of the process. This process is shown in Figure 2.9 (7,p.53), with Table 2.5 (7,p.53) listing AAG operating data. To date all tests have been done only at atmospheric pressure. Modifications have been made for accepting coal rather than char. The process would then be as shown in Figure 2.10 (7, p.52).

2.1.1.6 Advanced Gasification System (3.11.29.31)

A second generation fluidized bed process that may be desirable is that developed by the Westinghouse Electric Corporation. It seems to have the distinct advantage of being able to handle eastern caking coals. The process is shown in Figure 2.11 (11,p.87) with typical raw gas composition in Table 2.6 (11,p.88). Coal is blown through an open-ended concentric "draft tube" inside the gasifier vessel. The coal suspension plus the use of this concentric tube presumably keeps the coal from caking. In addition, at these high temperatures the ash agglomerates until they fall from the suspension. Caking deterrence and ash removal eradicates two major drawbacks of using eastern coals. Dolomite is used in the suspension for H_2S removal. Effective scale-up (in design now) would make this process very desirable and applicable in many situations.

2.1.2 High-Btu Gasification

High-Btu gasification produces a clean synthesized form of pipeline quality (950+ Btu/scf) "natural" gas. This gas could be used for industrial and residential consumption. The currently high operating costs of these rather complex processes make the gas relatively expensive. Coal feed and O_2 requirements are

common problems to most of the processes. These processes are almost exclusively second-generation fluidized and entrained-bed gasifiers. Indicating that the original emphasis in gasification technology was placed on high-Btu gas production.

2.1.2.1 BIGAS (1.3,4,5,9,11,13,31,48)

This technique uses a high-pressure, two-stage, gasifier as shown in Figure 2.12 (13,p.56). The reactor operates at rather extreme temperatures (2700°F), which may contribute to materials problems. The high pressures involved enhance the methanation process. The slagging removal system should be noted especially. No by-products are produced that require additional processing. Again, no tars or oils are produced.

This entrained bed system is said to have the advantage of being able to handle all types of coal. The overall process is shown in Figure 2.13 (11,p.75). However, a large quantity of fines are produced. This has prompted the performance of alternate coal preparation tests. These alternate paths are shown in Figure 2.14 (13, p.55).

Many of the problems still associated with this process, such as the coal preparation change mentioned above, are in the

pilot plant testing stage. Other tests involve water-cooling flows, materials testing, and ignition design. Operation of the process could be complex. The plant is essentially in the start-up phase; various pressurized tests are still required. This process seems to be best suited for eastern bituminous coals, although high ash fusion coals could be a problem for the slagging operation. Unfortunately, it has a large O_2 requirement, even though this requirement is reduced by preheating the fuel. Pressurized operation with air has not been demonstrated.

2.1.2.2 CO_2 Acceptor (1,3,4,5,9,11,31,34)

This process, shown in Figure 2.15 (11,p.76), is desirable because it does not require an external source of oxygen or hydrogen, reducing both initial and process costs. The H_2/CO ratio of purified gas is approximately 3:2, eliminating the need for shift conversion.

Materials handling and dolomite regeneration have caused some problems. In fact, operation of the process has proven quite delicate. However, the major drawback to this process is that it only operates effectively with the highly reactive lignite and sub-bituminous coals.

2.1.2.3 COGAS (9)

This conversion process is an outgrowth of the COED liquefaction process discussed below. It is being converted from char gasification to coal. Materials handling is one of the major operational problems involved.

2.1.2.4 HYGAS (1,3,4,5,9,11,14,15,16,31,33,51)

Rather extensive testing has been done on this process to determine its operating characteristics as a slurry-to-fluidized bed multi-stage gasifier technique with eventual scale-up to commercialization. All types of coal can be used, although it naturally operates best with highly reactive lignite. A 92 percent coal conversion has been attained with Illinois #6 coal. Mild-air oxidation pretreatment is required for certain types of coals in order to destroy the coal's agglomerating tendencies. The diagram in Figure 2.16 (15,p.75) illustrates the process.

The expensive aspect of this process is its need for a hydrogen-producing process (electrothermal, O_2 -steam, steam-iron). The process shown in Figure 1.17, being studied by IGT, utilizes the O_2 -steam generation system. There is less need for

methanation in this process compared to others. Table 2.7 (15,p.76) shows some of the operating characteristics and the relative raw gas yields.

This process is a representative example of those second generation processes where there might be a decision required between availability and cost. Tables 2.8 (33,p.29) and 2.9 (33,p.10) illustrate some of the costs associated with this process for a commercial 250 MMscfd plant using Pittsburgh seam coal.

2.1.2.5 Slagging Lurgi (3,10,52)

This process, as its name suggests, is an extension of the proven Lurgi technique. The major difference is the higher operating temperature (fusion temperature is restrictive in the original Lurgi process) for reducing present ash to slag for removal. This higher temperature improves the overall thermal efficiency of the process and, hopefully, will enable the process to handle high-agglomerating eastern coals. This has been demonstrated on a pilot plant scale (52). However, many of the problems inherent to the Lurgi process, such as scale-up, are still present in this, essentially first generation, process. Process comparisons with Lurgi are made in Table 2.10 (10,p.74).

2.1.2.6 SYNTHANE (1,3,4,5,6,9,11,18,31,32)

The Synthane process is a fluidized-bed gasification scheme being developed by the Bureau of Mines. If operated correctly, its inherent simplicity (no major recycle lines) could make it a favorable future technique. Figure 2.17 (18,p.81) is a detailed diagram of the procedure. External coal pretreatment is eliminated, and all types of coal should be applicable. Considerable methane is produced in the gasifier, which helps reduce methanation requirements. Table 2.11 (11,p.84) lists some typical raw gas compositions.

During testing there has been a problem with fines and clinkers. Oxygen requirements are still present, but are lower in relation to comparable processes. Materials testing is still required due to sulfur related corrosion problems. Many of these current problems should be resolved during pilot testing. Synthane would then be one of the more promising second generation techniques, with relatively inexpensive operating costs. Tables 2.12 (32,p.11) and 2.13 (32,p.24) list some representative capital and annual operating costs, respectively, for a hypothetical 250 MMscfd plant using Pittsburgh seam coal.

2.2 Coal Liquefaction

As stated above, the initial emphasis of coal conversion technology was aimed at developing a high-Btu gasification process, with low-Btu gasification applied to more localized operation. Indirectly, however, coal liquefaction has been drawing more and more support:

"But in the opinion of knowledgeable researchers in the government, the optimum path to be pursued is not one of coal gasification, nor even coal liquefaction, although for many reasons liquefaction is more attractive than gasification, but that of a multiproduct production In a plant of that type, the entire potential chemical energy of the coal would be employed usefully down to the last bit."
(9,p.275)

Coal liquefaction is a relatively new technology, although it was first developed in Germany in the 1920's and 1930's. One of the original techniques, Bergius, is no longer in operation anywhere in the world. The other, Fischer-Tropsch, has survived and is now used commercially (although not in the U.S.), most notably at the SASOL plant in South Africa. Some of the potential products of liquefaction by this process are exemplified in Table 2.14 (3,p.16). As can be seen, coal

liquefaction products could affect many markets, such as that of the automotive industry [37,38]. In fact, coal liquefaction's greatest advantage might be as a supplier of a synthetic crude oil:

"Production of synthetic crude allows it simply to be added to the natural crudes still available to refineries. With relatively minor modification to the refineries, final products essentially identical to present fuels result. This approach has the practical advantage of serving both the needs of oil companies wishing to maintain the usefulness of present investments and of insulating the consumer from change." [47,p.xii]

Coal liquefaction is accomplished by the addition of hydrogen (directly or indirectly) which results in the breakdown of the long and complex hydrocarbon molecular chains. Usually hydrogen is produced by a gasification or solvent process. Desulfurization usually occurs during a gasification step. There are four principal processes used to accomplish liquefaction (50.p.VII-5):

1. Hydrol liquefaction (direct catalytic hydrogenation)
2. Solvent extraction (necatalytic liquid phase-dissolution)
3. Pyrolysis

4. Liquid hydrocarbon catalytic synthesis (Fischer-Tropsch-indirect hydrogenation)

Hydroliquefaction and solvent extraction appear to be the most promising techniques for second generation processes. As the majority of the liquefaction processes use a form of hydrogenation, the primary differences among the processes occurs due to the amount of hydrogen required, catalyst requirements, and operational difficulties. These are all dependent on the type of multiproduct fuels demanded from, or available in, the coals used.

2.2.1 Fischer-Tropsch (3.10,19.20,50)

As in the Lurgi process, the current advantage of this process is its commercial availability. A typical line diagram for the process is shown in Figure 2.18 (11,p.77), with possible product composition given in Table 2.15 (11,p.78).

This technique uses an indirect liquefaction process, whereby coal is gasified to carbon monoxide and hydrogen, which are recombined catalytically to produce a variety of liquid and gaseous fuels. Various gasification processes could be used with this process although, if used in the U.S., it would be preferable to use a technique which can handle caking coals (at present those used do not). Another deterrent may be the

abundance of by-products unsuitable for fuel, such as waxes, although cleaner fuels are obtained here than with direct hydrogenation.

If used now, the economics of this process would compare favorably with direct hydrogenation processes. In the long run, the operation of a direct hydrogenation process would be cheaper. Thus, some feel the SASOL plant should be used as a model while others argue that these "antiquated" gasifiers should be ignored and the U.S. should concentrate on making available the more modern processes.

2.2.2 COED (1.3.9.11,50)

This process, shown in Figure 2.19 (11,p.77) liquifies the coal by first passing it through a series of pyrolyzing reactors. This was the first successful pilot process.

The residual char was first thought to make a good boiler fuel, but the demand for pipeline quality gas has led to development of a gasification process utilizing this char, with extensions to the use of coal. This outgrowth is the COGAS

process mentioned above.

This type of plant has been operated with Illinois, Colorado, and Wyoming coals. Temperature control in the various gasifiers is one of the major operating considerations. An oxygen plant is required, which is typical of most liquefaction processes. Table 2.16 (11,p.77) lists some representative yield data using Illinois #6 coal.

2.2.3 SRC (1,3,8,11,16,19,23,24,36,40,41,50)

Solvent refined coal is one of the original solvent extraction processes. It was originally developed for producing sulfur free coal. However, this is not its exclusive product.

"Only when product yields are placed in the perspective of everyday units of the using industry is it clear that the co-product streams are far from insignificant. The SRC process might more properly be called a multi-synthetic-fuels process." (21,p.89)

In fact, funding is currently being allocated to produce both liquid and solid SRC (22,p.7). It is hoped it can be ready for commercial application by 1987. The process line is shown in

Figure 2.20 (11,p.83).

In this process the coal is dissolved in a coal-based internally-generated solvent while adding hydrogen under pressure at a temperatures of 800°F. Hydrogen and other gases are released in the pressure letdown. If solid SRC is required, the slurry is cooled. Variations in hydrogen addition are required to produce more potential liquid fuels. It would seem that this slurry should be fed directly into a refining process; i.e., incorporate the process into a "coalfinery." This would appear to give a little more latitude in product composition. Some typical SRC process yields are shown in Tables 2.17 (21,p.69) and 2.18 (21,p.89).

An advantage of this process is the relative experience in using this technique. However, a drawback is the amount of hydrogen required. A potentially greater problem with the liquid fuel is meeting SO₂ emission requirements. Furthermore, this process does nothing to eliminate NO_x emissions. Tests are being done to study pollutant levels. A solvent deashing technique is being tested to remove sulfur and ash. Nevertheless, the many possible products from the SRC process may still make it one of the most desirable. Some representative costs are shown in Tables 2.19 (24,p.11) and 2.20 (24,p.34) for a 50,000 bpd SRC plant using Illinois #6 coal.

2.2.4 SYNTHOIL (3.8,11,36,50)

This is a hydrosulfurization process being developed to produce low ash, low sulfur fuel oil. The flow process is shown in Figure 2.21 (11,p.84).

As in many of these processes, coal is slurried with some of its own coal-based oil with the subsequent addition of more hydrogen at elevated pressure and temperature. It is then fed to a packed bed catalytic reactor. Typically, external hydrogen is required, but an additional catalyst (such as cobalt molybdate on silica-activated alumina) is required in the reactor.

2.2.5 Exxon Donor-Solvent (3.8,20,25)

A rapidly advancing technique is the donor-solvent process, in which coal is hydrogenated in a slurry generated by its coal-based solvent (oil). In this technique, however, the solvent is catalytically hydrogenated in a separate reactor before it is mixed with the coal feed. Hydrogen is supplied from the excess available in the donor solvent and from an external source. This may prove to be a very efficient process.

2.2.6 H-Coal Process (3.8,11,20,26,27,36,50)

The H-Coal process is a direct hydrogenation technique utilizing catalytic activation. The flow system is shown in Figure 2.22 (11,p.79).

This process is representative of the more advanced liquefaction technologies. An ebulliating bed of hydrogen, catalyst, coal, liquid and gaseous products is produced in the reactor and maintained at approximately 850°F and 2700 psig. Flash distillation at one atmosphere is used to separate light and heavy distillates. Typical products from Illinois No. 6 coal are shown in Table 2.21 (11,p.79).

The major drawback to the process is the extensive hydrogen requirement and, as in other processes, the cost of the catalyst. Tables 2.22 (27,p.12) and 2.23 (27,p.23) list some hypothetical costs for a 50,000 bpd H-Coal plant using Illinois coal.

2.3 Comments

It can be seen that liquefaction processes lean toward development of new techniques. This is also true for

gasification techniques, even though first generation techniques are being used in specific situations. The need for improvement is seen in the relative, and approximate, operating costs given in Table 2.15 (3,p.18).

Concerning the rating systems used in Tables 1.2-1.4, a shift is seen in emphasis from 1978 to the year 2000. In fact the "status rating" (SR) given relates to the present, whereas the "feasibility rating" (FR) is a rather subjective projection to the year 2000. The status rating is rather clear-cut and self-explanatory. It illustrates the state of the art. However, this could also be considered the feasibility rating for the year 1978, as the main criteria at this point in time is availability. There will be a steady shift in this feasibility rating in the upcoming years. By 1985, as the second generation processes become proven on a larger scale (and operating problems are resolved), it will be apparent that their availability is in the foreseeable future, and with more desirable characteristics.

Nevertheless a direction must be established for the aforementioned technologies. In selecting a frontrunner for the year 2000, alternative techniques are not discounted, but rather a configuration is established which will have the most probable significant impact on our fuel economy relative to its development. In doing so one can determine what hardware, and construction and operating costs, are required for a given flow

process. It should be pointed out that certain commercial applications are not advocated over others, but rather considered, reported, choices are made concerning the best overall technological process for the future. Upon establishing which concepts will predominate, the configurational and operational characteristics established by the appropriate group are used for an in-depth I-O cost analysis.

Consider Table 1.2. All technologies given in the table are worthy of consideration at present. The shift in ratings is indicative of the first and second generation processes. By 1985 there will be a phase-out of first generation processes, as second generation techniques gain economically and technologically. Note that modified first generation techniques will be in operation [43], but future construction plans should be shifting to newer processes. Lurgi will still have scale-up and caking coal problems, as will Wellman-Galusha. Koppers-Totzek will still have prohibitive O_2 requirements, and will probably still have fly-ash removal problems. This leaves the U-Gas and Winkler processes for the year 2000. Note that U-Gas is given a slightly higher rating. This is largely due to the fact that its testing has been the most extensive on a larger scale, and because of its use of the ash-agglomerating technique. Winkler is already a commercially proven process, and will feasibly be cost-competitive if O_2 is not used in the process. These processes are primarily desired for utility boiler

purposes, but they could also prove economically useful for a large industrial plant or an industrial park. Note that the future of low-Btu gasification technology lies with atmospheric fluidized bed processes. The development of U-Gas as a pressurized fluidized bed reactor process, resulting in increased efficiency, will make it the most desirable low-Btu gasification technique. This reactor is similar in concept to that of Westinghouse's Advanced Gasification System which was not discussed at length in this section simply because its development is in conjunction with a combined cycle system to be discussed later. Tables 2.24 (46,p.5-3) and 2.25 (46,p.5-2) list comparative costs for some general conceptual designs of second generation low-Btu gasification schemes for boiler application.

The transition in Table 1.3 is not as clear-cut simply because these are predominantly second generation gasification processes. However, those with the highest rating are those that have undergone sufficient testing to support optimism that the major problems with the process will be resolved or minimized. Again there is an elimination of some processes by 1985. The CO₂ Acceptor cannot be included because it may only be able to use lignite or sub-bituminous coal, which does not offer much of an environmental advantage and is not readily available in the ORBES region. In addition, slagging Lurgi will still have many of the problems inherent in the original Lurgi process. The COGAS process also will not be a factor past 1985, as it has materials

handling problems and will probably be used only as a char gasifier with its parent COED liquefaction process. BIGAS is also left out, primarily because of its extensive O₂ requirements. Methanation processes could possibly be available by 1985, but future plans will revolve around the remaining second generation techniques. These remaining processes for the year 2000 are Hygas and Synthane. Hygas has lower methanation step requirements, while the Synthane process has lower O₂ needs. In addition, no coal pretreatment is required for Synthane. Corrosion and fines problems are the major obstacles to be cleared. Note, that the most feasible processes utilize a fluidized bed reactor. However, while these seem to be the most economically and technologically attractive high-Btu processes, they are by no means inexpensive. Demand will ultimately dictate just how extensively they are used. For this reason the pipeline quality product gas will probably be used only for commercial/residential applications. The cost of production would seem to make it undesirable as a process fuel for industrial or utility uses. In fact, competition from low-Btu gasification and liquefaction processes may make its market too restrictive.

In Table 1.4 (liquefaction) some transition can be seen in the relative desirability of first and second generation processes. By 1985 Fischer-Tropsch, an indirect hydrogenation technique, will no longer be considered economically or

technologically competitive, as was the case with the original Bergius technique. The COED process will probably be shelved also as the pyrolysis method does not seem to be as efficient as the relatively newer methods. Thus, three desirable and increasingly popular [44] processes are left. The frontrunner at this time would seem to be SRC II. It is the most developed of the new techniques, and requires an internally-produced solvent. In addition two processes, H-Coal and Exxon Donor Solvent, are rated just behind in feasibility. This is mainly due to less development. H-Coal requires a catalyst, and Exxon, the support for which is increasing rapidly [42], requires both a catalyst and solvent. SRC is a solvent extraction technique, whereas H-Coal and Exxon are catalytic hydrogenation processes. Because all are multiproduct processes, the utilization is manifold. Even though hydrogen production is required for all processes, the various products make them desirable. In fact, coal liquefaction may have the best future of all coal technologies. These processes would seem to have the capability of providing commercially available help by the year 2000.

TABLE 2.1			
Raw Gas Compositions for LURGI			
Component	Mol %		
	High-BTU	Low-BTU	
CO	9.2	13.3	
CO ₂	14.7	13.3	
H ₂	20.1	19.6	
H ₂ O	50.2	10.1	
CH ₄	4.7	5.5	
C ₂ H ₆	0.5		
Other than H ₂ S	0.6	0.7	
N ₂		37.5	
Total	100.0	100.0	
Higher heating value (dry basis), BTU/scf	303	180	

Source: (10.p.80)

TABLE 2.2 Raw Gas Compositions for WELLMAN-GALUSHA			
Component	Mol %		
	High-BTU	Low-BTU	
CO	29.6	26.0	
CO ₂	12.3	3.0	
H ₂	30.3	13.9	
H ₂ O	25.3	8.3	
CH ₄	0.7	2.5	
N ₂	1.1	45.6	
O ₂	0.1		
H ₂ S/COS	0.6	0.7	
Total	100.0	100.0	
Higher heating value (dry basis), BTU/scf	268	168	

Source: (11,p.87)

TABLE 2.3 Raw Gas Compositions for KOPPERS-TOTZEK		
Component	Mol %	
CO	50.4	
CO ₂	5.6	
H ₂	33.1	
H ₂ O	9.6	
CH ₄	0.0	
H ₂ S+COS	0.3	
N ₂	1.0	
Total	100.00	
Higher heating value (dry basis), Btu/scf	180	

Source: (11,p.81)

TABLE 2.4			
Raw Gas Composition for WINKLER			
Component	High-BTU	Low-BTU	Mol%
CO	25.7	19.0	
CO ₂	15.8	6.2	
H ₂	32.2	11.7	
H ₂ O	23.1	11.5	
CH ₄	2.4	0.5	
H ₂	0.8	51.1	
H ₂ S	2,500 ppm	1,300 ppm	
COS	400 ppm	200 ppm	
Total	100.0	100.0	
Higher heating value (dry basis), BTU/scf	275	118	

Source: (11.p.88)

TABLE 2.5													
MAG Operating Data													
Run No. 1													
Number of cyclones													
Internal													
External													
Average bed temperature, °F.													
Coke breeze feed rate, lb/hr													
Average solids feed mesh size													
Venturi throat diameter, in.													
Venturi throat velocity, ft/sec.													
Total bed superficial velocity, ft/sec.													
Product gas heating value, Btu/scf													
Carbon gasification efficiency, %													
Finest elutriation rate, lb/hr													
Appropriate production rate, lb/hr													
Appropriate mesh size													
Average appropriate mesh size													
Product gas composition (dry)%, %													
CO ₂													
CO													
H ₂													
CH ₄													
H ₂													
Calculated higher heating value based on gas analysis, Btu/scf													
Collected in first- and second-stage external cyclones, whether fines recycle was used or not.													
Fines collected with appropriate cyclones, whether fines recycle was used or not.													
Source: (7.p.53)													

TABLE 2.7

HYGAS Pilot Plant Operating Characteristics

	Period 1	Period 2
Reactor operating conditions:		
Coal or char feed (dry), lb/hr	5,912	5,478
Process steam feed, lb/hr	6,117	6,894
Process oxygen feed, lb/hr	824	960
Reactor pressure, psig	1,010	893
Steam-oxygen zone temperature, °F.	1,640	1,556
Operating results		
Raw gas composition (dry basis), mole %		
H ₂	22.73	26.65
CO ₂	34.30	35.30
C ₂ H ₆	0.47	1.39
N ₂	6.68	4.44
H ₂ S	0.70	0
CH ₄	26.48	23.31
CO	8.63	8.91
Total	100.00	100.00
Methane yield, scf/lb coal	2.8	4.1
CARBON OXIDE YIELD, scf/lb coal	4.5	7.7
Coal gasified, %	53.0	74.0
Carbon gasified, %	45.0	67.0

*Period 1--Test 54, period 7/10/76 (0000 hr) to 7/11/76 (0000 hr); feed pretreated char from Illinois No 6 bituminous coal.

TABLE 2.6

Raw Gas Composition for "Advanced Gasification" Process

Component	Mol %
CO	17.7
CO ₂	8.6
H ₂	13.3
H ₂ O	7.9
CH ₄	2.5
N ₂	50.0
H ₂ S	small quantity
Total	100.0

Higher heating value (dry basis), 156 Btu/scf

Source: (11,p 88)

Unit	Cost	Percent	Estimated number of operators per shift
Coal preparation	\$3,270,200	0.7	2
Coal pretreatment	16,577,000	3.3	2
Char slurry preparation	5,111,800	1.1	2
Hydrogenation	78,413,500	15.7	4
Dust removal	3,497,700	.7	2
Waste heat recovery No. 1	5,195,600	1.1	1
Shift conversion	5,076,400	1.0	1/2
Waste heat recovery No. 2	16,195,500	3.2	2
E-T-X recovery	6,043,100	1.2	2
Purification	61,510,000	12.3	5
Methanation	5,451,100	1.1	2
Waste heat recovery No. 3	5,428,100	1.1	1-1/2
Oxygen plant	29,164,000	5.8	2
Sulfur recovery	6,687,000	1.4	3
Waste water treatment	14,013,500	2.8	2
Flue gas processing	20,141,200	4.0	2
Drying	910,600	.2	1
Steam and power plant	41,093,000	8.2	4
Plant facilities	24,547,400	4.9	6
Plant utilities	35,184,700	7.1	4
Total construction	307,031,400	77.5	50
Initial catalyst requirements	7,455,100	1.5	
Total plant cost (insurance and tax bases)	394,486,500	79.0	
Interest during construction	59,173,000	11.9	
Subtotal for depreciation	453,659,500	90.9	
Working capital	85,366,000	9.1	
Total investment	\$599,025,500	100.0	

Source: (33,p.29)

	Unit Cost	Total	Percent
Direct cost:			
Raw materials and utilities:			
Coal	665.7 tph x 7,920 hr/yr x \$11/ton	\$57,995,800	
Raw water	1,200 M gph x 7,920 hr/yr x \$0.15/M gal	1,425,600	
Catalyst and chemicals		3,405,600	
Methane	77.5 M scfm x 7,920 hr/yr x \$0.75/M scf	46,400	
Direct labor:			
1,200 man-hr/day	\$6/man-hr x 365 day/yr	2,621,000	
Supervision	15% of labor	394,200	
Plant maintenance:			
487 men	\$15,000/yr	7,305,000	
Supervision	20% of maintenance labor	1,461,000	
Material and contract	150% of maintenance labor	10,957,500	
Payroll overhead	30% of payroll	19,723,500	16.0
Operating supplies	20% of plant maintenance	3,456,500	2.9
Total direct cost		93,517,300	75.9
Indirect cost (administration and general overhead)			
	40% labor, maintenance, and supplies		
Fixed cost:			
Taxes and insurance	2% of plant cost		
Depreciation	5% of subtotal for depreciation		
Total operating cost, before credits		134,766,200	108.9
Credits:			
Sulfur	165.04 tpd x 330 day/yr x \$25/ton	1,520,850	1.2
Ammonia	133.42 tpd x 330 day/yr x \$100/ton	2,041,700	2.1
B-L-X	63,211 tpd x 330 day/yr x \$0.25/gal	6,569,500	5.6
Operating cost, after credits		123,726,000	100.0

Coal at \$13 per ton, operating cost = \$134,727,700.
 Coal at \$15 per ton, operating cost = \$144,617,400
 Source: (33,p.10)

TABLE 2.10				
A Process Data Comparison of Normal and Slagging LURCI Gasifiers				
Item	Units	Fixed-bed Processes		
		Process 11	LURCI Dry-Ash	LURCI Slagging
Coal (4.7% Moisture)	ton/day	12,860		13,410
Oxygen	ton/day	5,700		8,050
Cooling Water Circulated	gpm	263,000		356,000
Boiler Feed Water Makeup	gpm	5,300		3,900
Fired Boiler Steam Production	M lb/hr	1,470		693
Total Steam Production	M lb/hr	4,765		3,103
Dry Gas to CO Shift	MM scfd	310		785
Dry Raw Synthesis Gas	MM scfd	883		802
Methane by Synthesis	MM scfd	133		166
CO ₂ Removal	ton/day	16,050		18,500
Sulfur	ton/yr	49,000		59,000
Ammonium Sulfate	ton/yr	187,500		187,500
Raw Phenols	ton/yr	20,500		15,200
Char	ton/day	---		---
Overall Efficiency BTU in Gas/BTU in coal	Percent	66.9		64.1

Source: (10, p.74)

TABLE 2.11			
Raw Gas Composition for Synthane Process			
Components	Mol %		
	High-BTU	Low-BTU	
CO	10.5	5.4	
CO ₂	18.2	14.9	
H ₂	17.5	14.3	
H ₂ O	37.1	23.8	
CH ₄	15.4	7.0	
C ₂ H ₆	0.5	0.3	
C ₃ H ₈		0.1	
H ₂ S	0.3	0.3	
N ₂	0.5	33.9	
Total	100.0	100.0	
Higher heating value (dry basis), BTU/scf	405	186	

Source: (11, p.84)

TABLE 2.12 Total Estimated Capital Requirements and Operating Personnel for SINTHANE Plant			
Unit	Cost	Percent	Estimated number of operators per shift
Coal Preparation	\$15,610,300	3.00	3
Gasification	83,062,300	15.98	7
Dust removal	17,282,200	3.32	2
Shift conversion	7,674,300	1.48	2
Waste heat recovery	13,263,200	2.55	1
Purification	53,950,800	10.37	4
Methanation	45,167,000	8.68	5
Pipeline compression	1,416,000	.27	1
Final drying	1,165,500	.22	1
Oxygen plant	37,050,000	7.12	2
Sulfur recovery plant	1,875,000	.36	2
Waste water treatment	16,356,000	3.14	2
Steam and powerplant	52,667,100	10.12	6
Plant facilities	25,990,600	5.00	4
Plant utilities	37,253,400	7.16	4
Total construction	409,787,900	78.77	46
Initial catalyst requirements	1,405,500	.28	
Total plant cost (insurance and tax bases)	\$411,253,400	79.05	
Interest during construction	61,688,000	11.86	
Subtotal for depreciation	472,941,400	90.91	
Working capital	87,208,100	9.09	
Total investment	\$560,235,500	100.00	

Source: (32,p.11)

TABLE 2.13 Estimated Annual Operating Cost for SINTHANE Plant			
	Unit Cost	Total	Percent
Direct costs:			
Raw materials and utilities:			
Process coal	593.7 tph x 7,920 hr/yr x \$ 11/ton	\$51,723,100	
Utility coal	15.3 tph x 7,920 hr/yr x \$11/ton	1,332,900	
Raw water	1,020 M gal/hr x 7,920 hr/yr x \$0.15/M gal	1,211,600	
Catalyst and chemicals		1,707,200	
Direct labor:			
1,104 man-hr/day	16/man-hr x 365 day/yr	2,447,600	
Supervision	15% of labor	362,100	
Plant maintenance:			
500 men	\$15,000/yr	7,620,000	
Supervision	20% of maintenance labor	1,524,000	
Material and contracts		11,430,000	
Payroll overhead	30% of payroll	3,577,400	
Operating supplies	20% of plant maintenance	4,114,600	
Total direct cost		87,021,700	71.6
Indirect cost (administration and general overhead)	40% of labor, maintenance, and supplies	10,987,700	9.1
Taxes and insurance	2% of plant cost	8,225,100	6.8
Depreciation	5% of subtotal for depreciation	23,607,100	19.5
Total operating cost, before credits		129,881,600	107.2
Credits:			
Salvage	270.8 tph x 330 day/yr x \$25/ton	1,137,400	1.4
Labor	4,849 mph x 7,920 hr/yr x \$0.10/mph	3,840,400	3.2
Amortiza	159.84 tph x 330 day/yr x \$10/ton	1,168,700	2.6
Operating cost, after credit		121,139,000	100.0

Operating cost, coal at \$13 per ton = \$107,085,000.
Operating cost, coal at \$15 per ton = \$140,432,200.

Source: (32,p.28)

TABLE 2.14
SASOL Liquefaction Products

Gasoline	Styrene	Ammonia
Fuel Oil	Higher Alcohols	Ammonia nitrate
Diesel Oil	Acetone	Ammonia Sulfate
Town Gas	Ketones	Crude Phenols
LPG	Xylol	Motor Benzole
Methanol	Naphtha	Benzene
Motor Alcohol	Waxes	Toluol
Ethanol	Paraffin	Road Tar Prime
Ethylene	Creosote	Electricity
Butadiene	Pitch	

Source: (3,p.16)

TABLE 2.15

Product Composition of FISHER-TROPSCH

	Composition (vol%)	
	Fixed-bed Process	Fluid-bed Process
Liquidified petroleum gas (C ₃ --C ₄)	5.6	7.7
Petrol (C ₅ --C ₁₁)	33.4	72.3
Middle oils (diesel, furnace, etc.)	16.6	3.4
Waxy oil or Gatsch	10.3	3.0
Medium wax, mp 203°-206°F.	11.8	
Hard wax, mp 203°-206°F.	11.8	
Alcohols and ketones	4.3	12.6
Organic acids	traces	1.0

	Liquid-product Composition (vol%)	
	Fixed-bed Process	Fluid-bed Process
Paraffins	45	55
Olefins	50	40
Aromatics	0	0
Alcohols	5	5
Carbonyls	traces	traces

Source: (10,p.78)

TABLE 2.16

Yield Data for COED Pyrolysis
of Illinois #6 Coal

Net Process Yield	Wt. % of Dry Coal
117 lb/ton	59.1
1.04 bbl/ton	19.6
7.1 gal/ton	5.5
18.133 scf/ton	15.8
Total	100.00

Source: (11,p.77)

TABLE 2.17

Typical SRC Yield

	Wt %	
	Raw Coal	Product Solvent Refined Coal
Carbon	70.7	88.2
Hydrogen	4.7	5.2
Nitrogen	1.1	1.5
Sulfur	3.4	1.2
Oxygen	10.3	3.4
Ash	7.1	0.5
Moisture	2.7	--
	100.0	100.0
Volatile matter	38.7	36.5
Fixed carbon	51.5	63.0
Ash	7.1	0.5
Moisture	2.7	--
	100.0	100.0
BTU/lb	12,821	15,768

Source: (21,p.89)

TABLE 2.18		
SRC Gas and Liquid Yields		
C ₁ -C ₄ gas, scf		3,130
CH ₄ , scf		2,100
C ₅ -350°F., gal		32
bbl		0.762
350°-750°F. distillate, gal		38
	bbl	0.904
Total liquid,	gas	70
	bbl	1.666

*Per ton of solvent refined coal from hydroliquefaction reactor. †Approximate analysis of C₁-C₄ gas cut:

	Vol %	Value/cu ft	BTU
CH ₄	67.0		680
C ₂ H ₆	19.3		340
C ₃ H ₈	10.0		260
C ₄ H ₁₀	3.7		120
Total	100.0		1,400

Source: (21,p.89)

TABLE 2.19			
Total Estimated Capital Requirements and Operating Personnel for SRC Plant			
Unit	Cost	Percent	Estimated Number of Operators Per Shift
Coal preparation	\$15,426,200	2.2	5
Coal Slurrying and Pumping	1,502,500	.2	1/2
Coal liquefaction and filtration	151,261,000	21.9	5
Dissolver acid gas removal	55,507,000	7.5	1-1/2
Coal liquefaction and product distillation	6,096,600	1.1	1-1/2
Fuel oil hydrogenation	60,458,900	8.6	2-1/2
Naphtha hydrogenation	5,307,000	.8	1/2
Fuel gas sulfur removal	4,423,500	.6	1-1/2
Gasification	19,145,400	2.6	1-1/2
Acid gas removal	20,622,500	3.0	1-1/2
Shift conversion	16,459,000	2.4	1
CO ₂ removal	9,751,000	1.4	1
Methanation	759,300	.1	1
Sulfur recovery	4,200,000	.6	1-1/2
Oxygen plant	25,000,000	3.6	1-1/2
Product storage and slaw removal	15,994,700	2.3	1
Steam and powerplant	48,604,400	7.0	3
Process waste water treatment	3,513,300	.5	1/2
Plant facilities	34,960,100	5.0	2
Plant utilities	50,109,400	7.1	2-1/2
Total construction	551,263,600	76.7	36
Initial catalyst require-ments	2,669,300	.4	
Total plant cost (insurance and tax bases)	553,872,900	79.1	
Interest during construction	63,060,400	11.6	
Subtotal for depreciation	636,553,600	90.9	
Working capital	63,555,400	9.1	
Total investment	700,649,200	100.0	

Source: (24,p.11)

TABLE 2.20			
Estimated Annual Operating Cost for EAC Plant			
	Unit cost	Total	Percent
Direct cost:			
Raw materials and utilities:			
Coal	$11 \text{ t/ton} \times 20.456 \text{ t/d} \times 330 \text{ d/yr}$	774,255,260	
Raw water	$15 \text{ M gpm} \times 80.157 \text{ gal} \times 1,440 \text{ min/day} \times 330 \text{ d/yr}$	1,069,200	
Calorific and chemicals		5,307,600	
Direct labor:			
664 man-hr/day	$86 \text{ man-hr} \times 365 \text{ day/yr}$	1,092,200	
Supervision	15% of labor	283,600	
Plant maintenance:			
664 men	$15,000/\text{yr}$	10,760,000	
Supervision		2,052,000	
Material		15,390,000	
Payroll overhead	30% of payroll	27,762,000	16.2
Operating supplies	20% of plant maintenance	4,386,400	2.5
Total Direct Cost		120,396,600	70.6
Indirect cost (administration and general overhead)			
803 labor, maintenance, and supplies		18,167,400	8.3
Fixed cost:			
Taxes and insurance	2% of plant cost	11,017,500	6.5
Depreciation	5% of subtotal for depreciation	31,847,700	18.7
Total operating cost, before credits		177,889,400	108.1
Credits:			
Sulfur	$634 \text{ tpd} \times \$25/\text{ton} \times 330 \text{ day/yr}$	5,230,500	3.1
Power	$22,000 \text{ kW-hr/hr} \times \$0.010/\text{kW-hr} \times 330 \text{ day/yr} \times 24 \text{ hr day}$	1,742,400	1.0
Operating cost, after credits		170,516,500	100.0

Coal at \$13 per ton, operating cost = \$184,017,500.
Coal at \$15 per ton, operating cost = 197,516,500.

Source: (24.p.34)

TABLE 2.21		
H-COAL Products from Illinois No 6 Coal		
IBP cuts, °F.	Vol %	Gravity °API
IBP - 400	42.18	44.6
400-650	41.51	17.3
650-975	16.31	5.0
Total	100.00	25.2

Source: (11.p.79)

TABLE 2.22 Total Estimated Capital Requirements and Operating Personnel for H-COAL Plant			
Unit	Cost	Percent	Estimated Number of Operators Per Shift
Coal preparation	\$28,442,800	4.3	8
Hydrogenation	215,615,100	32.7	8
Refinery gas cleanup	15,941,200	2.4	1
Oxygen plant	31,000,000	4.7	2
Hydrogen production	61,289,000	9.3	6
Hydrogen compression	18,489,600	2.8	3
Ammonia and hydrogen sulfide removal	2,688,200	0.4	2
Sulfur recovery	4,500,000	0.7	1
Tankage	21,835,100	3.3	2
Steam and powerplant	38,332,100	5.8	3
Plant facilities	32,860,000	5.0	5
Plant utilities	47,099,300	7.2	3
Total construction	518,092,400	78.6	44
Initial catalyst requirements	3,086,900	0.5	
Total plant cost (insurance and tax bases)	521,179,300	79.1	
Interest during construction	78,176,900	11.8	
Subtotal for depreciation	599,356,200	90.9	
Working capital	59,935,600	9.1	
Total investment	659,291,800	100.0	

Source: (27,p.12)

TABLE 2.23 Estimated Annual Operating Cost for H-COAL Plant			
	Unit Cost	Total	Percent
Direct costs:			
Raw materials and utilities:			
Coal	23,512 ton/day x 330 day/yr x \$11/ton	\$85,348,600	
Water	660 Mghp x 24 hr/day x 330 day/yr x \$0.15/M gal	784,100	
Catalyst and chemicals		13,755,200	54.4
Direct labor:			
Supervision	1,056 man-hr/day		
Supervision	86 man-hr x 365 day/yr	2,312,600	
Plant maintenance:			
443 men	\$15,000/yr	9,645,000	
Supervision	205 of maintenance labor	1,929,000	
Material and contracts		14,467,500	
Payroll overhead	30% of payroll	4,270,000	2.3
Operating supplies	20% of plant maintenance	5,208,300	2.8
Total direct cost		133,417,900	79.1
Indirect cost (administration and general overhead)	40% labor, maintenance, and supplies	13,563,700	7.4
Fixed costs:			
Depreciation	2% of plant cost	10,423,600	5.7
Interest and insurance	5% of Subtotal for depreciation	29,907,300	16.3
Total operating costs, before credits		192,022,300	104.5
Credits:			
Amortization	120.6 short ton/day x 330 day/yr x \$60/ton	2,383,600	1.3
Sulfur	710.6 short ton/day x 330 day/yr x \$25/ton	5,662,400	3.2
Operating cost after credits		183,776,700	100.0

Operating cost with coal at \$13/ton = \$199,293,900.
Operating cost with coal at \$15/ton = \$214,011,800.

Source: (27,p.23)

Plant Section	Case 1	Case 2	Case 3
Coal Preparation	2,850	2,228	2,228
Oxygen Supplies	7,064	5,992	1,774
Gasification	5,426	7,016	7,216
Compression, Gas Expansion	1,706	1,758	3,350
Desulfurization and Dehydration	889	904	1,622
Sulfur Recovery	1,332	2,701	1,780
Interconnecting Piping	1,825	1,353	1,492
Utilities	478	565	466
Direct Field Cost	21,994	21,941	20,372
Distributable Field Cost	2,195	1,352	1,628
Total Field Cost	24,189	23,293	21,000
Engineers, Home Office, and Fee	2,661	2,562	2,420
Total Construction Costs	26,850	25,855	24,420
Startup Costs	2,950	2,844	2,686
Total Capital Costs	29,804	28,699	27,106
Basis			
Blast Mode	O ₂ /steam	O ₂ /steam	Air/steam
Gasifier Type	Entrained solids	Fluidized solids	Fluidized solids
Sulfur Content in Coal, %	2	2	2
Sulfur Emission in Gases, e (SO ₂) lb SO ₂ per 10 ⁶ Btu HHV of coal	1.2	1.2	1.2
Gas Treatment Pressure, optimum, psia	150	150	150

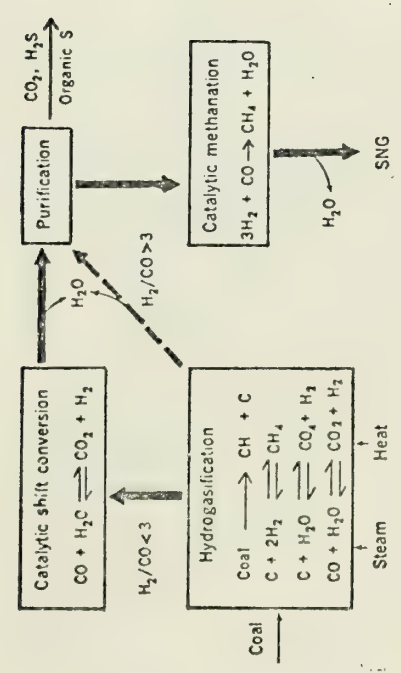
Source: (47,p.5-3)

TABLE 2.25

Cost Element	Case 1	Case 2	Case 3
Coal @ \$25/ton	2,969	2,889	3,313
Purchased electric Power @ \$.030/kw-hr	673	736	245
Catalysts and Chemicals	80	80	80
Equipment, Supplies, Utilities	80	80	80
Operating Personnel	528	528	528
Maintenance Materials and Labor	800	800	800
Total Annual Operating Costs	5,130	5,113	5,046
Basis			
Blast Mode			
Gasifier Type	O ₂ /steam	O ₂ steam	Air/steam
	Entrained solids	Fluidized solids	Fluidized solids
Sulfur content in coal, %	2	2	2
Sulfur Emission in Gases lb SO ₂ per 10 ⁶ Btu HHV of Coal	1.2	1.2	1.2
Gas Treatment Pressure Optimum, psia	150	150	150

Source: (47,p.5-2)

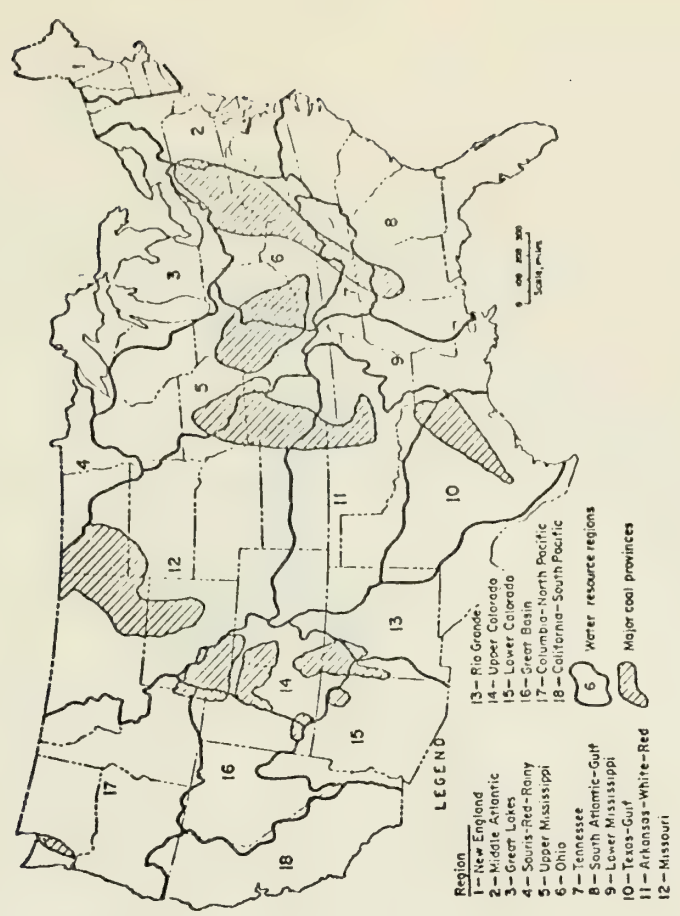
FIGURE 2.1



Generalized Gasification Flow Chart

Source: (5,p.45)

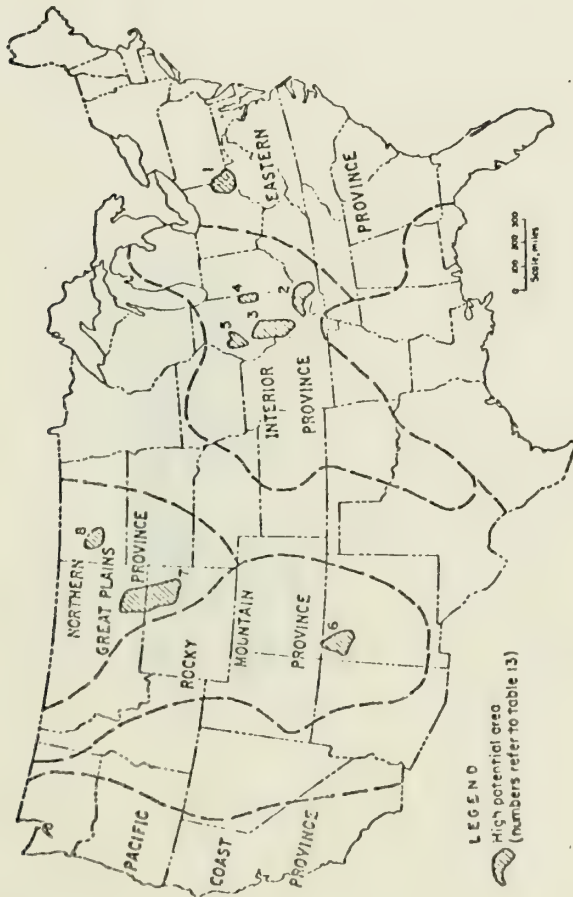
FIGURE 2.2



Major Water and Coal Resource Regions

Source: (6,p.27)

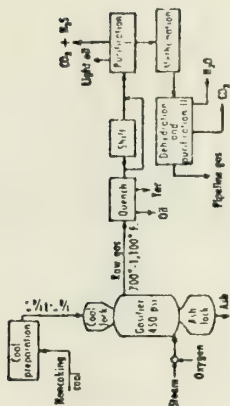
FIGURE 2.3



Areas of High Potential for Gasification Development

Source: (6,p.33)

FIGURE 2.4



LURGI Process

Source: (11,p.80)

FIGURE 2.5

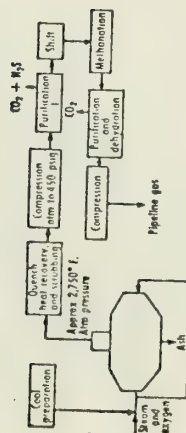


FIGURE 2.6

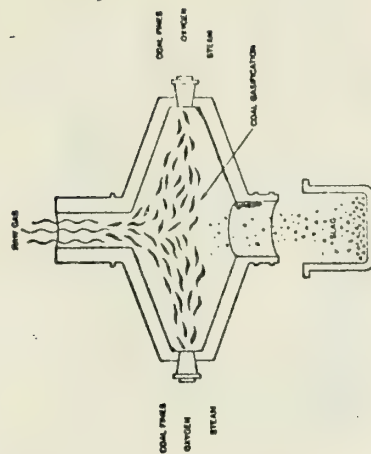
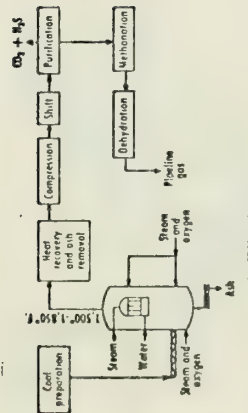


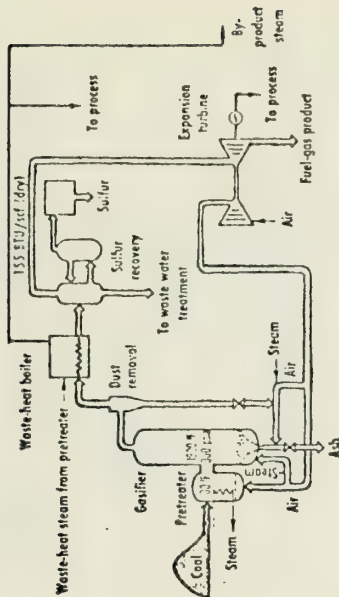
FIGURE 2.7



WINKLER Process

Source: (11,p.88)

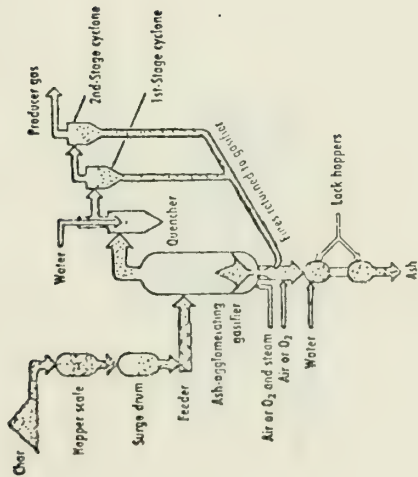
FIGURE 2.8



U-GAS Process

Source: (7,p.51)

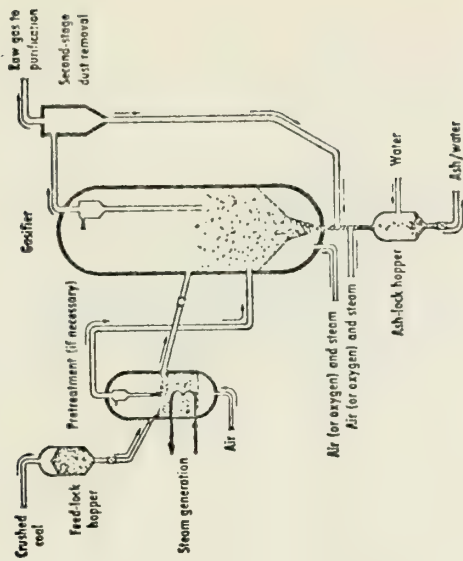
FIGURE 2.9



AAG Pilot Unit

Source: (7.p.53)

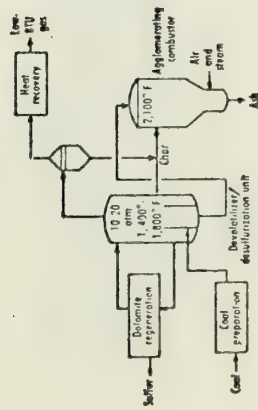
FIGURE 2.10



Modified U-Gas Process

Source (7.p.52)

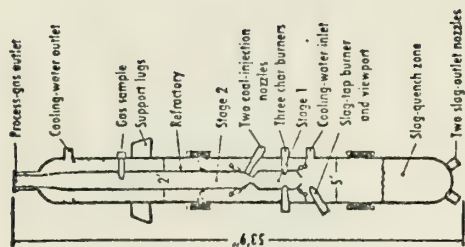
FIGURE 2.11



Westinghouse Coal Gasification Process

Source: (11,p.87)

FIGURE 2.12



Pilot Plant BI-GAS Gasifier

Source: (13,p.56)

FIGURE 2.13

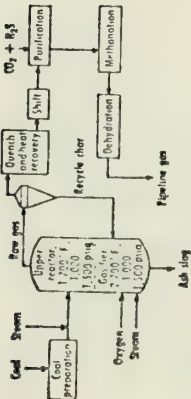


FIGURE 2.14

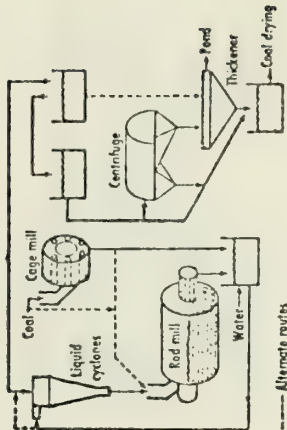
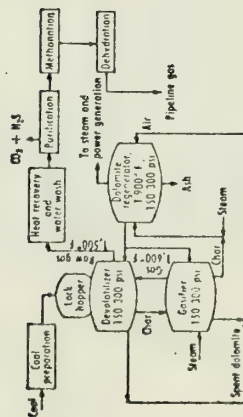
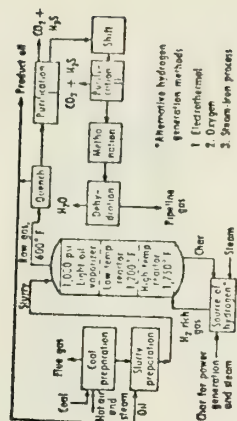


FIGURE 2.15



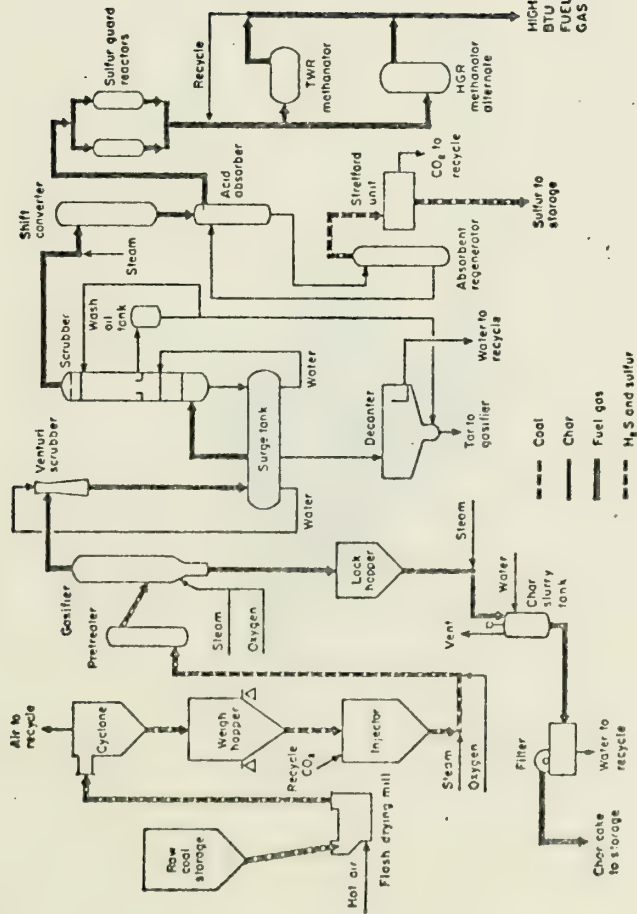
Source: (11.p.76)

FIGURE 2.16



Source: (15, p.75)

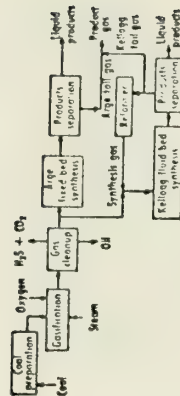
FIGURE 2.17



SYNTHANE Plant Flow

Source: (18,p.281)

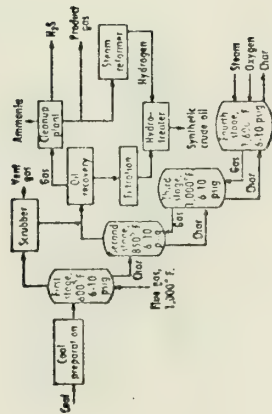
FIGURE 2.18



FISCHER-TROPSCH Process

Source: (11,p.77)

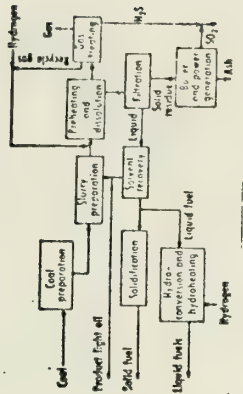
FIGURE 2.19



COED PROCESS

Source: (11,p.77)

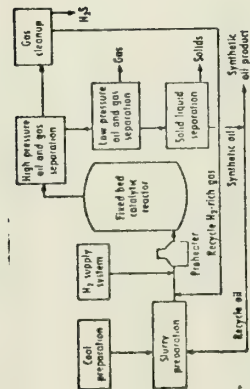
FIGURE 2.20



SRC PROCESS

Source: (11,p.83)

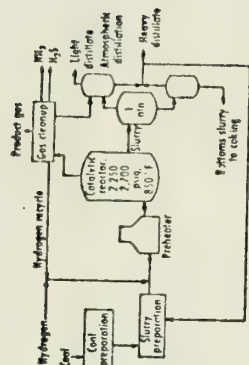
FIGURE 2.21



SYNTHOIL PROCESS

Source: (11, p. 84)

FIGURE 2.22



H-COAL PROCESS

Source: (11, p. 79)

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3.0 COMBINED CYCLE

Among the routes considered for improved coal utilization are advanced power cycles. The primary advantage is increased fuel cycle efficiency, resulting in fewer pollutants for given levels of generated power [2]. The leading future alternatives are: 1) high temperature gas-steam turbine cycles. air-cooled turbine with hot-gas clean-up, 2) high-temperature gas-steam turbine cycle, ceramic turbine elements, utilizing pre-cleaned coal, and 3) advanced steam plants utilizing pressurized fluidized bed combustion (FBC) with in-bed sulfur removal (4). Major future improvements under this regime involve developing combined cycle configurations. Combined cycle processes are being studied by Westinghouse (7) and the Bureau of Mines (6). The Westinghouse system, in conjunction with its aforementioned Advanced Gasification System, appears to be the most advanced and economical process.

Figure 3.1 (1,p.2-8) illustrates a typical combined cycle process flow. A conventional combined cycle system involves a low-Btu gasification process followed by cleanup of the product gas before it is burned in a gas and steam turbine system (combined cycle). Note the waste heat recovery line from the gas turbine system to the steam boiler. This has been used effectively in many industrial applications (8). This overall

concept appears to have the lowest reported capital cost (1,p.2-20).

The parameters which will determine the future of this process are coal gasification advancement, environmental restrictions, and gas turbine technology. Future advantages include better resource utilization, lower natural gas requirements, reduced emissions, and higher thermal efficiency. Table 3.1 (1,p.2-2) compares some of the characteristics of a combined cycle system utilizing the COGAS process with those of a pulverized coal boiler with a tail-gas scrubber. Note that these are, as yet, only potential advantages. Extensive study has been conducted (4) for this process using Westinghouse's Advanced Gasification System (see Section 1). It is, however, not currently operational.

The advantages cited are dependent upon improvements in coal gasification techniques. This implies advances in second generation gasification techniques so that ORBES region coals can be used; eliminating the cost of excess coal transportation. That is, the cost-competitiveness of the entire system must justify increases in thermal efficiency. Improved thermal efficiency goes beyond process design for a combined cycle system. Improvements in gasification techniques not only result in better utilization of the coal, but create the possibility of higher temperatures for use in the system. However, this

requires the development of a high temperature gas clean-up system (primarily for H₂S) in order to retain this thermal potential. Gas turbine blades that can withstand increased temperatures are also required. Table 3.2 (1,p.2-23) lists the effects on efficiency of these improvements. Note that here the gasifier is not assumed to affect efficiency, but it will affect cost appreciably. The base case represents expected future improvements.

The thermal improvement capability of a combined cycle system is assumed to be an increase in efficiency from 36 to 44 percent (1). However, this assumes solution of all technological (and relevant economic) questions. This is required in order to offset the high investment costs. The time horizon for combined cycles is well into the future. For example, the discussion here assumes improved coal gasification techniques and a high temperature process in which the thermal potential is retained by the future development of a hot-gas cleanup system. Hot-gas cleanup is currently possible only for streams with very few tars and oils. In general, a commercial hot-gas cleanup process will not be available until at least 1985 (1,p.2-18), as the one closest to development is the rather expensive iron oxide process. A dolomite-based hot cleanup process (9) also seems favorable to alternative wet gas cleanup techniques, but this has

been proven only on a bench scale.

High temperature handling is desired so that the turbine gas inlet temperature will be higher. Table 3.3 (1,p.2-22) illustrates some desirable operating specifications. However, note that high turbine gas inlet temperatures are not seen to be feasible until 1995. This is due to the need to improve the metallurgical characteristics of the gas turbine blades in order to cope with these higher temperatures. At present, 2000°F is the maximum allowable temperature. Improved blade materials (such as ceramics) are also needed in order to withstand possible damage from fines entrained in the hot fuel gas stream. Cyclones appear to be the particular removal system least affected by high pressure and temperature, but this rather inexpensive system has a particulate size limitation of about 5 microns or larger (1,p.2-16).

Clearly, many of the component problems in a combined cycle system place the earliest available date of an effective commercial process well into the 1990's. Studies are still required concerning instrumentation, multiphase flow characteristics, combustion characteristics, and erosion control [5]. This does not include the integration of these components into a single operating unit. In addition, the high temperature conditions increase NO_x emission from the gas turbine; an increasingly important environmental consideration.

Studies (3) would also seem to indicate that hot gas cleanup systems with second generation gasification schemes may not offer the large thermal and economic incentives initially anticipated. Tables 3.4 (3,p.5-8) and 3.5 (3,p.5-9) list comparative capital costs for combined cycle systems at low and high temperatures, respectively. This analysis was done for certain reactor characteristics and gas cleanup systems, the characteristics of which are defined in Table 3.6 (3,p.4-7). A similar comparison of operating costs is given in Tables 3.7 (3,p.5-10) and 3.8 (3,p.5-11).

Because there are characteristic start-up and turn-down problems it appears that this system would have to be used as a base-load operation. It does not now appear that the tradeoff between potential operating advantages and subsequent required improved technology of the high-temperature combined cycle system will allow this process to be commercially effective within the desired ORBES time horizon. This goes beyond the dependence on the development of a second generation gasification technique and subsequent environmental control process. Table 3.9 (10,p.35) compares the status of various advanced power systems.

TABLE 3.1 Potential Environmental Impact of 1000 MW Power Plant		
	Pulverized Coal Boiler With Stack Gas Scrubber	Gasification- Combined Cycle Power Plant
Coal Consumption: LB/KWH	0.82	0.70
Limestone Required: LB/KWH	0.15-0.20	0
NO _x Emissions: PPM	500	400
SO ₂ Emissions: PPM	200	10-200
Particulate Emissions: GR/SCF	0.01	Comparable
Cooling Tower Heat Rejection: BTU/KWH	4,200	2,500
Makeup Water: GAL/KWH	0.6-0.65	0.4-0.45
Disposal Land Required: Acres/1,000 MW	1,200-2,400	200
Projected Maximum Efficiency (1990)	36%	44%

Source: (1,p.2-2)

TABLE 3.2 Typical Efficiency Considerations and Effects			
	Base Case	Revised Design Condition	Change In Efficiency %
1. Turbine Inlet Temperature		2600°F	2000°F -8.0%
2. Gas Cleanup Temperature		100°F	800°F +1.0%
3. Gasifier Pressure		300 psi	atmospheric -1.0%
4. Gasifier Type		Second Generation	First Generation 0%
5. Air vs. Oxygen Blown		Air	Oxygen -1.0%

Source: (1,p.2-23)

TABLE 3.3	
Combined Cycle Specifications	
Steam Plant:	2400 psi, 1000°F, 1000°F (reheat), heat rate = 9000 Btu/kwhr, unfired waste heat recovery system
Gas Turbine:	Inlet temperature 2000°F to 2800°F (1975) (1995)
Combined Cycle:	Capacity Factor: base load = 0.70 Intermediate Load = 0.45 Power Split: 58% gas turbine, 42% steam turbine

Source: (1.p.2-22)

Summary of Estimated Component Capital Costs 1,950° Gas Turbine Inlet Temperature											
Case	MAL	MAL	MAL	MAL	MAL	MAL	MAL	MAL	MAL	MAL	FXH
Output MW (100%)	1,000	1,000	1,000	1,000	1,000	1,000	1,000	1,000	1,000	1,000	1,000
Cost 1 M lb/hr	950.4	787.1	899.6	753.8	764.4	743.4	768.0	762.9	734.1	725.7	
Coal Preheating	14.169	11.751	13.415	11.254	11.412	11.038	11.764	11.389	10.960	10.834	
Gas Cooling/Heating	53.793	1.543	34.215	5.166	7.269	4.194	24.436	6.065	26.555	7.400	
Oxygen Unit	63.644	63.934	---	---	56.322	53.719	70.956	74.704	---	---	
Air Compression	---	26.443	---	---	---	---	---	---	26.712	28.391	
Water Treatment	19.636	---	75.682	---	26.602	---	---	---	1.400	---	
Cold Purification (1)	37.640	---	47.358	---	26.753	---	31.313	---	38.003	---	
Hot Purification	---	73.021	---	72.968	---	52.344	---	56.146	---	---	
Combined Cycle	176.900	200.000	178.146	200.000	200.000	200.000	200.000	200.000	200.000	200.000	
Utilities	109.204	70.480	97.389	63.182	61.383	60.095	63.225	64.059	56.467	58.173	
Tar fired Power Unit	115.656	---	109.369	---	---	---	---	---	---	---	
Subtotal	726.996	478.656	649.878	431.817	415.665	406.974	433.057	439.305	400.514	396.438	
Continuity	10.050	21.198	97.480	64.772	62.350	61.046	64.954	65.905	60.077	59.766	
Plant Investment	836.050	500.454	747.358	496.589	478.015	468.020	498.016	505.270	460.541	458.204	
Income Taxes	20.901	13.761	16.684	12.415	11.950	11.701	12.450	12.632	11.515	11.455	
Freight/Production Costs	48.421	32.022	43.391	28.963	27.971	27.440	29.129	29.472	26.951	26.601	
Royalties	20.801	13.761	18.684	12.415	11.950	11.701	12.450	12.632	11.515	11.455	
Capital & Contingents	4.160	2.752	3.737	2.483	2.340	2.490	2.490	2.526	2.303	2.291	
Construction Interest	145.168	94.570	128.874	85.632	82.474	80.785	85.818	87.129	79.424	79.011	
Total Recoverable Capital	1,014.821	707.612	900.134	634.516	614.712	601.447	640.414	649.660	532.299	529.218	
Working Capital	82.948	30.654	39.104	24.570	28.057	27.300	28.102	28.945	26.945	26.781	
Total Capital	1,117.769	738.266	939.238	659.086	642.769	628.746	668.516	678.605	559.244	556.000	
Net Total Capital	---	---	---	---	---	---	---	---	---	---	
Net Total Investment	---	---	---	---	---	---	---	---	---	---	

Source: (3.p.5-8)

Summary of Estimated Development Costs and Time to Operation of Demonstration Plants									
Millions of Mid-1975 Dollars					Millions of Escalated Dollars				
System	Time to Operation of Demonstration Plant (years)	Add Costs Through Pilot Plant Operation	Demonstration Plant Costs	Total	Add Costs Through Pilot Plant Operation	Demonstration Plant Costs	Total	Comments	
Advanced Steam-ATB	6	---	129	---	---	340	---	Includes ATB and PFB for demonstration but not development costs.	---
Closed cycle gas turbine-helium Uranium bottoming	15	200	647	847	266	1373	1661	Includes ATB for pilot and demonstration plant but not development costs.	1661
Potassium topping cycle	16	308	266	594	572	701	1273	Includes PFB for pilot and demonstration plant but not development costs.	1273
Open cycle gas turbine-combined-water cooled- demulson liquid fuel	16	150	142	292	207	300	507	Includes fuel processing development costs.	507
Open cycle gas turbine-Ltlu Gasifier	11	46	137	103	55	233	288	Includes gasifier for demonstration plant but not development costs.	288
Open cycle MHD	19	---	---	---	---	---	---	not development costs.	3074

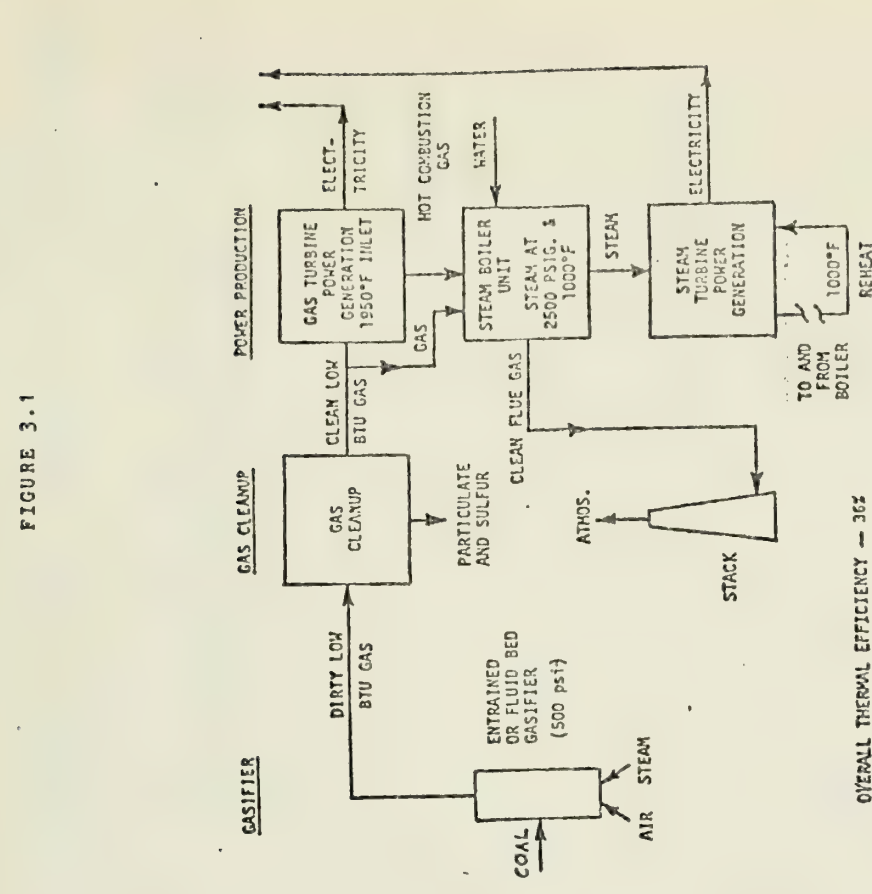
TABLE 3.1

Estimated Cost of Electrical Power for a 2,400⁰g Gas Turbine Inlet Temperature and a 70 Percent Load Factor

Case	HL	HXL	HXH	NAL	NAL	NAL	NSL	MSL	ESL	EXH	EXH	EXH
Output (70%), Mw	700	700	700	700	700	700	700	700	700	700	700	700
Unit rate, \$/kwh	10.544	6.556	9.907	6.265	6.624	6.409	9.028	6.666	6.359	6.215	597.02	597.02
Total Capital Requirement, \$M	1,046.443	702.67	935.933	641.65	629.125	605.5	657.65	603.90	603.90	597.02	597.02	597.02
Operating charges, \$/MWh	64.656	52.478	60.750	50.604	52.642	51.504	55.360	51.275	51.275	50.374	50.374	50.374
Local at \$1/H \$/M	6.616	5.768	7.700	5.263	5.151	4.956	5.366	5.267	4.946	4.890	4.890	4.890
Operator Labor	0.546	0.367	0.540	0.355	0.326	0.316	0.343	0.343	0.343	0.315	0.311	0.311
Chemical and Chemicals	12.923	6.653	11.550	7.695	7.727	7.437	6.062	6.416	6.060	7.335	7.335	7.335
Maintenance Labor	16.014	12.061	16.100	11.005	10.771	10.367	11.266	11.264	10.341	10.224	10.224	10.224
Material	6.462	4.326	5.775	3.948	3.663	3.719	4.041	4.040	3.709	3.667	3.667	3.667
Administrative and Support Labor	15.665	10.466	14.000	9.570	9.366	9.015	9.797	9.795	6.992	6.890	6.890	6.890
General and Admin. Expense	19.561	13.110	17.447	11.962	11.707	11.269	12.246	12.243	11.250	11.113	11.113	11.113
As before taxes and insurance	0.159	0.225	0.062	0.140	0.035	0.028	0.062	0.062	0.023	0.016	0.016	0.016
Utilities	0.254	0.206	0.236	0.199	0.208	0.202	0.217	0.209	0.201	0.196	0.196	0.196
Capital charges, \$/MWh	40.266	26.943	35.990	24.605	24.054	23.163	25.193	25.195	23.143	22.662	22.662	22.662
Depreciation	21.347	16.534	19.043	13.093	12.634	12.352	13.420	14.407	12.320	12.179	12.179	12.179
Avg. fund interest, %	32.021	21.402	26.640	19.660	19.751	18.578	20.130	20.110	18.479	18.269	18.269	18.269
Avg. return on equity, %	34.679	25.266	31.017	21.770	20.649	20.066	21.601	21.760	20.013	19.765	19.765	19.765
Avg. income tax, %	275.191	193.769	246.932	179.730	174.066	173.005	167.426	165.100	172.376	170.115	170.115	170.115
Cost of Capital, \$/MWh	44.4	27.6	40.6	29.3	29.2	28.2	30.6	30.2	28.1	27.7	27.7	27.7
Cost of Power, \$/MWh	55.7	40.4	50.7	37.7	36.0	36.6	39.8	39.1	36.7	36.1	36.1	36.1
Cost of Power, \$/MWh	55.7	40.4	50.7	37.7	36.0	36.6	39.8	39.1	36.7	36.1	36.1	36.1

Summary of Estimated Development Costs and Time to Operation of Demonstration Plant									
System	Time to Operation of Plant (years)	Millions of Mid-1975 Dollars			Millions of Escalated Dollars			Comments	
		R&D Costs Through Pilot Plant Operation	Demonstration Plant Costs	Total	R&D Costs Through Pilot Plant Operation	Demonstration Plant Costs	Total		
Advanced Steam-ATR	6	---	129	---	183	---	---	Includes ATR and FR for demonstration but not development costs.	
Advanced Steam-ATR	12	---	160	---	340	---	---	Includes ATR for pilot and demonstration but not development costs.	
Closed Cycle Gas Turbine- Helium Organic Bottoming	15	200	647	847	1373	1661	---	Includes ATR for pilot and demonstration plant but not development costs.	
Potassium topping Cycle	18	308	286	594	701	1273	---	Includes FR for pilot and demonstration plant but not development costs.	
Open Cycle Gas Turbine- Combined-Water Cooled- Semitrlean Liquid Fuel	14	150	142	292	300	507	---	Excludes fuel processing development costs.	
Open Cycle Gas Turbine Lbtr Gasifier	11	46	137	183	233	268	---	Includes gasifier for demonstration plant but not development costs.	
Open Cycle RHD	19	561	893	1444	2249	3074	---	Includes gasifier for demonstration plant but not development costs.	

Source: (10,p.35)



Conventional Combined Cycle Process

Source: (1,p.2-8)

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4.0 DIRECT COAL COMBUSTION

4.1 Flue Gas Desulfurization

This section considers the state-of-the-art of flue gas desulfurization (FGD) techniques. FGD is used almost exclusively by the electric power industry. Alternatives to an FGD system include coal precleaning, SRC, FBC, or combined cycle coal gasification. While some FGD systems are time-proven, none are without undesirable features: 1) cumbersome and costly equipment for relatively small SO₂ gas volumes, 2) corrosion and erosion problems, and 3) difficulty with power plant integration due to their chemical nature. Some of the methods are expensive, still need improvement, or simply cannot meet the standards set by the 1977 Clean Air Act Amendments.

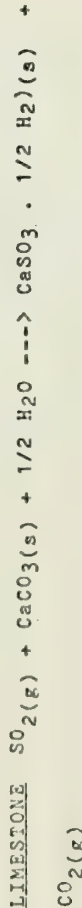
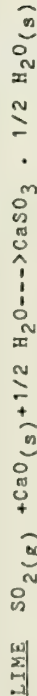
While FGD systems remain the most desirable cleaning process, as measured by utility acceptance, possible improvements provide a basis for the segregation of these techniques. First generation systems can be termed throwaway, while second generation techniques are regenerative. These terms refer to how the waste products are handled; the former refers to the disposal of the SO₂ and its absorbent, while the latter indicates recycling the absorbent and possible utilization of the wastes in

the form of some usable product. Table 4.1 lists the leading processes under their respective headings. There are obviously many more processes than those listed. However, those processes listed are the most advanced. Improvements in these will more than likely produce improvements in others. Table 4.2 categorizes the relevant information for each process. A chart formulated by Stukel (4.p.11), was used for much of the information. The listing of Status Ratings (SR) and Feasability Ratings (FR) are defined in the same way as those used for the coal gasification and liquefaction processes.

FGD is the primary emission control process used at this time. However, this discussion is not an attempt to compare throwaway and regenerative techniques. Regeneration does have potential advantages. It also has higher costs. Rather, a discussion of each process is developed with an examination of the main problems and costs of these processes. As lime/limestone is the most widely used currently available technique, the major portion of the discussion deals with this process. The comparable merits and disadvantages of the most feasible new techniques comprise the remainder of this section.

4.1.1 Lime-Limestone Wet Scrubbing (1.2.3.4)

This process is the most effective and, hence, most widely used to date. Of the 31 operating systems, 26 use this technique (2,p.11). The ratio will probably increase. The technique involves using a slurry of calcium oxide or calcium carbonate to absorb SO₂ in a wet scrubber, with the resulting calcium sulfite and sulfate disposed of as waste solids. A flow diagram of the process is shown in Figure 4.1 (1,p.337). The overall absorption reactions are shown below:



The main problems associated with this process are (2,p.11): SO₂ removal efficiency, wet-dry interface deposition, scaling, mist elimination, gas reheat, and corrosion-erosion. Future tests are required to determine the trade-offs in dealing with these problems. Even after an effective solution, the inherent problems of waste disposal and transportation of lime or

limestone remain.

Process efficiency depends on the amount and type of lime or limestone used for scrubbing. Figure 4.2 (2,p.36) shows a large available deposits in the ORSES region. The type of scrubber contacts is also very important for efficient absorption. Characteristics of the more popular types are shown in Table 4.3 (1,p.351).

The main problem with this technique is sludge disposal. This is substantial in both weight and volume, requiring a large waste pond or landfill. Table 4.4 (1,p.348) lists the waste product composition for this process. The environmental effects for disposal alternatives are listed in Table 4.5 (5,p.34), with some comparative cost ranges given in Table 4.6 (5,p.32).

A breakdown of the mass and energy requirements are given in Tables 4.7 (1,p.344) and 4.8 (1,p.345), for a typical limestone and lime scrubbing operation, respectively. Representative investment and operating costs are given in Tables 4.9 (2,p.56) and 4.10 (2,p.58) respectively. Tables 4.11 (3,p.55) and 4.12 (3,p.56) also compare some typical costs to those of other processes. Possible improvements (2,p.61) in the process may be made by the use of mass transfer additives, forced oxidation to improve sludge dewatering and scaling reduction, and sparged scrubbing.

4.1.2 Double Alkali (2,3,4)

This technique is actually a hybrid of the throwaway and regenerative processes. Flue gas SO_2 is removed by contact with a soluble alkali salt (sodium, potassium, or ammonia). This solution is then treated with another alkali element (calcium) which regenerates the salt and removes a calcium-sulfur solid. The sludge produced is similar to that in the lime-limestone process. This technique has the advantages of scale-free scrubber operation and high SO_2 removal. If operated properly, it should have no corrosion problems. Stukel (4,p.22) illustrates the process and lists the relevant overall reactions.

The major drawbacks at this time are the possibility of water pollution if not handled properly, the occurrence of scaling if not operated properly, and that the system requires an external energy source, which reduces overall plant efficiency. Another problem is that the present process can use only lime which, because it requires a kiln operation, is more expensive and less energy efficient than limestone. The future of this process seems to depend on the operating characteristics of a large scale plant. The capital and operating costs in Table 4.9 and 4.10 should be noted.

4.1.3 Sodium Scrubbing (1,2,3,4)

This technique involves the absorption of SO_2 in a sodium sulfite/bisulfite solution. A simplified process diagram, with reactions, is shown by Stukel (4,p.31). A detailed diagram illustrating the equipment is shown in Figure 4.3 (1,p.247). The spent scrubbing solution is processed to regenerate the scrubbing solution and separate the absorbed SO_2 . The SO_2 is then processed into a final product for marketing. The obvious advantages are a reusable scrubber and a marketable product to reduce costs. A key advantage is that the scrubbing process is proven and well-defined.

The Wellman-Lord process uses the thermal regeneration technique for recovering the scrubbing solution. This simply involves reversing the absorption reaction by the addition of heat. The product is a concentrated SO_2 stream that can then be converted into elemental sulfur or sulfuric acid. The primary disadvantage is that this process may be very expensive. It is energy intensive, and a significant amount of sodium sulfate is produced which must be removed. The major drawback may be that currently natural gas is used as a reducing agent (for SO_2 conversion). Typical mass and energy requirements are given in Table 4.13 (1,p.256). At this time it seems the future of the

process lies in small industrial applications.

A process variation, developed by Atomics International, may be more promising. This is another sodium scrubbing operation, but sodium carbonate is used as the absorbent. The major advantage is that coal (or coke) may be used as the reducing agent for the solids. This yields sodium sulfide which is converted to H_2S and, ultimately, elemental sulfur. Tests are still being done on the overall process.

4.1.4 Magnesia Scrubbing (1,2,3,4)

This technique is probably the simplest and most promising of the advanced processes. Its main advantages are the ease of absorbent regeneration, the ability to separate the power and chemical units, and the minimal effect on the power plant due to the elimination of a solids disposal problem. A magnesia-based slurry is used to absorb SO_2 to form magnesium-sulfur compounds. The scrubbing agent is then regenerated while the recovered SO_2 is processed to form liquid SO_2 , sulfuric acid, or elemental sulfur. The primary absorbent reaction is:



The magnesium oxide is regenerated by thermal decomposition:

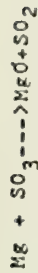


Figure 4.4 (1,p.311) illustrates a typical magnesia slurry scrubbing process. A magnesia scrubbing process can use a magnesia base slurry, solution, or solid, but the slurry system is the most advanced.

Current estimated costs make this system competitive, and there do not seem to be any undesirable products. However, due to the relatively low SO_2 concentration in the kiln offgas, the primary conversion process being used yields sulfuric acid rather than elemental sulfur. This acid is not as good a product as elemental sulfur as it is impractical to store for extended periods of time. In addition, the market is relatively small. This process is also somewhat energy intensive, and there are questions concerning the regenerative capabilities of magnesia.

However, the Chemico-Basic magnesia slurry SO_2 recovery system for control of sulfur oxides has been demonstrated successfully on an oil-fired boiler at Boston Edison, and has been tested on a coal-fired boiler at Potomac Electric [6]. The system showed the ability to remove 90% of the SO_2 present. Particulate control capability of the system was also demonstrated. Magnesia was regenerated and recycled successfully, and commercial grade sulfuric acid was produced and marketed. Long duration runs are still required for the system

because major shutdowns occurred due to corrosion problems.

4.1.5 Comments

Currently the lime/limestone slurring technique is the proven process. It will remain so until newer processes are refined. The magnesia-based system seems to be the most desirable regenerative technique. Future pollution regulations could dictate an even larger use, and development, of FGD systems. A possible enhanced advantage for regenerative techniques is related to a potentially massive sludge disposable problem which could be offset by the much smaller volumes of marketable or disposal sulfur. Other techniques, such as catalytic oxidation, have also shown desirable features, but at this point it seems they are still at least a decade away. In any case, the requirements of raw materials, water, energy and manpower, will be constant determinants. This includes requirements for scrubber back-up systems and maintenance. In fact, as indicated by future construction plans, it appears that the first generation lime/limestone scrubber will be the primary process for quite some time.

However, it must be kept in mind that just as EPA standards encouraged the use of FGD systems, the standards could also change to influence the position of the throwaway systems. That

is, EPA regulations imposed on throwaway landfills or ponds could make the system costs prohibitive. The desirability of the regenerative systems would then be enhanced. Refinement of the regenerative processes would then be accelerated. The by-product sulfur market would then compete with Frasch process sulfur.

4.2 Fluidized Bed Combustion

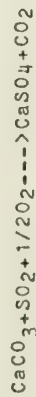
Fluidized bed coal combustion (7,8,9) is of major technological interest due to the advent of strict environmental restrictions on combustion processes and the increasing costs of clean-burning oil and gas. There are three regimes of sulfur removal, a primary criterion, for coal combustion:

- 1 Removal of sulfur before combustion-coal precleaning
2. Removal of sulfur oxides after combustion-flue gas desulfurization (FGD).
3. Removal of sulfur compounds during combustion-fluidized bed combustion (FBC)

This third is of concern here. Because coal precleaning will not have a significant commercial impact even with improving technology before 1985 (7 p.5-1), comparisons made here will be

between FBC and FGD systems.

Fluidized bed combustion involves a mass of particles undergoing combustion while held in suspension by an upward flow of air. This suspension is in contrast to other modes of coal feed: entrained and moving beds. Carbon particles in the flue gas are collected and recycled for further burning. The ideal particle size for the system is 1/4 to 3/4 inch maximum (8,p 5) because this eliminates the need for pulverization equipment. This size seems to burn quickly and completely. Desulfurization is accomplished by the use of limestone which is fed directly into the bed. The primary reaction for sulfur removal is shown below (8,p.4):



The main components of an FBC unit operation are: 1) reaction vessel, 2) solids feed or flow control, 3) solids discharge, 4) dust separator, 5) instrumentation, and 6) gas supply (12)

The potential advantages of an FBC system over conventional combustion with FGD are offered below (7,p.5-2).

1. Lower coal preparation cost
2. The ability to use coals with high ash content
3. Higher heat transfer - less heat transfer surface

required

4. Higher heat generation - smaller boilers
5. Reduced corrosion due to lower temperature
6. Lower SO₂ emission
7. Lower NO_x emission
8. The possibility of passing the stack gas through a gas turbine for increased efficiency

However, additional work is required to determine the optimum characteristics for various aspects of the process (9). Scale-up is required to determine the optimum bed design for the system. This includes the use of limestone for SO₂ control. (Like most FGD systems, limestone supply and throwaway are additional cost considerations.) More research is needed in order to understand the heat transfer characteristics and to improve the overall combustion efficiency. Further study is also needed to determine the erosion limits of the materials used. Although at this time NO_x and SO₂ standards are met, a reduction in these standards could become a factor, limiting the use of FBC systems. NO_x emission could also increase with increased coal combustion (due to reaction with fuel-bound nitrogen) or from changes in

operating temperature.

There are two types of FBC systems - atmospheric and pressurized. An atmospheric system could be used for a conventional utility boiler steam cycle. A typical system is shown in Figure 4.5 (7,p.5-7). Quenching caused by the boiler tubes becomes an additional design consideration. This boiler would replace the conventional pulverized or cyclone fired boiler. Fly ash is removed by electrostatic precipitation, and air heaters for recovery are included, as well as most of the conventional auxiliaries. Monongahela Power Company has installed and is testing (11,p.2) a 30-Mw atmospheric demonstration boiler at Rivesville, West Virginia (8,p.5). This is a Foster-Wheeler product.

Pressurized (approximately 10 atm.) FBC systems would be used primarily with a combined gas/steam turbine cycle. Here the combustor would be used as a supercharged boiler between the air compressor and the gas turbine. A typical system is shown in Figure 4.6 (7,p.5-8). Coal dust combustion is directly controlled by the amount of air supplied to the fuel. Some of the attractive features of a pressurized FBC system include (7,p.5-9):

1. Increased combustion rates due to lower temperature-less carbon carryover
2. Higher steam production with smaller bed size

increase

3. Lower NO_x emissions

4. Coal to electric conversions as high as 45 percent-reduced capital investment over atmospheric FBC (possibly 20 percent)

One of the limitations of this system is the temperature. Above 1900°F there is a shift in the chemical equilibrium for sulfur removal and free SO₂ increases. In addition turbine blade materials cannot tolerate temperatures much greater than this. There are also design problems, because of excessive corrosion of boiler tubes due to particle motion and turbine blade damage due to particulates. These have been seen experimentally as damaging possibilities. A pilot plant using pressurized FBC is planned by the Curtiss - Wright Company at Woodbridge, New Jersey (8,p.6). This 13-MW plant is scheduled for start-up in 1980.

In addition to the temperature limitations mentioned above, both types of FBC systems must maintain a temperature of at least 1400°F. Below this the combustion efficiency is too low, the SO₂ removal reaction is too slow, and the heat transfer advantages are lost. Thus the operating temperature range is quite restrictive. In addition turndown is, at most, 50 percent of full load (and usually 75 to 80 percent (7,p.5-5)), restricting its use for variable load operation. The start-up of an FBC system is also quite complex, and will probably require some type

of external firing.

Despite the shortcomings mentioned above, a comparative cost analysis (7) was done for conventional combustion systems with FGD, and atmospheric and pressurized FBC systems. The cost estimates assume some resolution of operating problems, but at best are only of comparative interest. The results for capital and operating costs are shown in Table 4.14 (7, p.5-13) and Table 4.15 (7, p.5-14), respectively. However, 80% sulfur removal has been assumed for the FBC systems, which is now unacceptable. This would cause an increase in operating costs, if an increase in removal is possible at all.

If the design and operating problems mentioned above can be resolved, Tables 4.14 and 4.15 demonstrate a possible advantage for FBC systems. However, at this time it appears that while atmospheric FBC systems might be applicable in specific situations within the near future, pressurized FBC systems will probably not be commercially available until the 1990's at best. A possible use for an atmospheric FBC system might be as a base load operation for a power plant, since startup and shutdown seem to be major operational problems. Even here coal type (e.g. sulfur content) could be a limiting factor. Another possible utilization is industrial process heating, such as in a refinery or other cogeneration units:

"But industry, where nearly one third of all U.S. energy is consumed, was not receiving attention. Primarily for this reason, the Energy Research and Development Administration (ERDA) issued contracts to develop FBC's for industrial uses.

"An evaluation of FBC's for refinery and chemical-plant fired process heaters is under way at Exxon Research and Engineering Co. the goal of Exxon's program is to make a complete technical and economic assessment of the FBC concept." (10, p.96)

Industrial utilization may be a more feasible long-term application for all FBC systems rather than competing with conventional utility techniques. Fluidized bed systems have already been proven for industrial processes involving chemical reactions, drying, heating, or cooling in various stages of product manufacture (12). Thus application of coal combustion in a fluidized state would seem to be a viable alternative for industries in or out of the ORBES region.

TABLE 4.2						
Status of FGD Systems						
FR	SK	Process Name	Developed By	Process Remarks	Status Remarks	Fuel Type Eff.
1	1	Limestone Scrubbing	---	Throwaway Process; Sludge Disposal Problem	Commercially Available	Coal 70-85
4	1	Lime Scrubbing	---	Throwaway Process; Sludge Disposal Problem	Commercially Available	Coal 60-95
3	2	Double Alkali Scrubbing	---	Throwaway/Regenerative Process; Scaling and Corrosion Problems	Commercially Available (Construction)	Coal 90-95
4	2	Sodium Scrubbing Thermal Regeneration	Wellman-Lord	Regenerative Process; Elemental Sulfur or Sulfuric Acid Product; Requires Natural Gas For Reduction	Commercially Available (Construction)	Coal ---
3	3	Sodium Scrubbing Melt Reduction	Atomics International	Regenerative Process Uses Coal as Reducing Agent; Elemental Sulfur Product	DEMO. (1LST)	Coal 60-90
2	3	Magnesia Scrubbing	EPA, Philadelphia Electric	Regenerative Process; Sulfuric Acid Product, Simple Process	DEMO. (1LST)	Oil 90

TABLE 4.1	
THROWAWAY	REGENERATIVE
Wet Lime/Limestone	Sodium Scrubbing - Thermal Regeneration
Double Alkali	Sodium Scrubbing - Melt Reduction Magnesia Scrubbing

TABLE 4.3						
COMPARISON OF SCRUBBER TYPES FOR A LIMESTONE WET SCRUBBING SYSTEM						
Parameter	Scrubber Type					
	Marble Bed	TCA	Venturi	Grid Tower	Spray Tower	
SO ₂ Removal Efficiency	Good	Good	Fair	Good	Good	
Particulate Removal Efficiency	Good	Good	Excellent	Good	Fair	
Typical L/G (gal/1000 scf) for SO ₂ Removal	40-70	50-85	20-50	50-100	70-110	
Gas Side Pressure Drop (in H ₂ O) for L/G Above	8-12	6-12	8-20	1-7	1-3	
Gas Velocity (ft/sec)	3-8	6-11	125-300	6-11	5-25	
Dissolution of Solids	Good	Fair	Poor	Fair	Poor	
Resistance to Solids Plugging	Fair	Good	Excellent	Fair	Excellent	

Source: (1.p.351)

TABLE 4.4					
TYPICAL COMPOSITIONS AND QUANTITIES OF WASTE PRODUCT FROM LIME/LIMESTONE FGD SYSTEMS					
Lime Waste (dry)	Weight Percent	Production (tons/yr)	Assumed Packing Volume (ft ³ /ton) of waste	Approximate Volume Required For Storage in 30 Years (acre-feet)	
CaSO ₃ ·1/2H ₂ O	60	65,741			
CaSO ₃ ·2H ₂ O	27	38,099			
CaCO ₃	11	14,657			
Inerts	2	3,051			
TOTAL	100	141,748	22	2.148	
Lime waste Sludge (wet, 50% solids)		263,496	45	4.393	
Limestone Waste (dry)					
CaSO ₃ ·1/2H ₂ O	53	65,741			
CaSO ₃ ·2H ₂ O	24	38,099			
CaSO ₃	15	23,835			
Inerts	8	12,493			
TOTAL	100	160,168	22	2.427	
Limestone Waste (wet, 50% solid)		320,336	45	4.964	

Assumptions:

Coal: 3.5% S
Plant: 500 MW, 5,260 hr/yr, 375,000 lb/hr coal
Scrubber: 90% SO₂ removal
86% utilization of CaO
79% utilization of CaCO₃

Source: (1.p.348)

TABLE 4.5 ENVIRONMENTAL EFFECTS OF DISPOSAL ALTERNATIVES					
Type of Disposal	Condition of Waste	Primary Drainage	Environmental Effects		
Pond	Untreated ^a	Supernate	Seepage	Runoff	Land Reuse
	or chemically fixed ^b	Supernate	Yes	No	No
Basin	Untreated ^a	Underdrainage	Yes	No	No
	or conditioned ^c				
Landfill	Conditioned ^c	Runoff	Yes	Yes	Yes
	or chemically fixed ^b				

^aUntreated waste refers to FGD sludges as emitted from primary or secondary dewatering equipment.

^bChemically fixed sludges refer to the waste treated by one of several commercial processes that make these wastes suitable for landfill disposal.

^cConditioned waste refers to sludge treated by techniques other than chemical fixation and includes oxidation to gypsum and dewatering by mixing with dry fly ash or other agents that allow the material to be handled in a manner similar to that for soils.

Source: (5,p.34)

TABLE 4.6 SLUDGE DISPOSAL COST RANGES (UNTREATED AND CHEMICALLY TREATED LANDFILLS, 1000-MW STATION, 50 PERCENT LOAD FACTOR, 30-YEAR AVERAGE, JANUARY 1976 DOLLARS)					
Disposal Method	Base Material	S/Ton Sludge ^{a,b} (Dry)	\$/Ton Coal ^b	Mills/kwh ^{b,c,d}	
Untreated Pond	Natural Clay ^e	3.50	1.00	0.4	
	Linert ^f	5.70-7.60	1.60-2.20	0.7-1.0	
Chemically Treated ^g	Indigenous Soil	7.30-11.40	2.10-3.20	0.9-1.4	

^a510,000 short tons/year average (dry basis) including fly ash.

^bCoal burned at rate of 0.88 lb/kwh. 3% sulfur, 12% ash, 85% SO₂ removal, 1.2 CaCO₃/SO₂ mole ratio.

^cLand costs at \$1000/acre are included (equivalent to \$0.25/ton sludge, dry).

^dDisposal within 5 miles of power plant.

^eAssumes coefficient of permeability of clay is 1 x 10⁻⁶ cm/sec or better.

^fPonding costs cover range based on low-to-high material costs. i.e., PVC-20 (low) to Hypalon-30(high).

^gChemical fixation costs vary, depending on characteristics of the waste and the disposal process chosen.

Source: (5,p.32)

TABLE 4.7					
Raw Material and Energy Consumption for Limestone Scrubbing					
Basis: 500 Mw, 3.5 Percent Sulfur Coal, 5,260 hr/yr					
Process Area	Limestone tons/hr	Steam MM Btu/hr	Water M gal/hr	Electricity Mw	
SO ₂ Absorption	25	77	25	10.1	
Solids Separation	-	-	-	0.1	
Solids Disposal	-	-	11	0.1	
TOTAL	25	77	36	10.3	

Source: (1,p.344)

TABLE 4.8					
Raw Material and Energy Consumption for Lime Scrubbing					
Basis: 500 Mw, 3.5 Percent Sulfur Coal, 5,6260 hr/yr					
Process Area	Lime tons/hr	Steam MM Btu/hr	Water M gal/hr	Electricity Mw	
SO ₂ Absorption	11.6	77	25	9.5	
Solids Separation	-	-	-	0.1	
Solids Disposal	-	-	10	0.1	
TOTAL	11.6	77	35	9.7	

Source: (1,p.345)

TABLE 4.9

LIMESTONE SLURRY PROCESS
SUMMARY OF ESTIMATED FIXED INVESTMENT^a

(500-Mw new coal-fired power units, 3.55 S in coal; 1.2 lb SO₂/Mbtu heat input allowable emission; on-site solids disposal)

Direct Investment	Investment, \$	Percent of total direct investment
MATERIALS HANDLING (conveyors, feeders, conveyors, elevators, bins, stackers, miller)	1,155,000	6.0
Feed preparation (feeders, crushers, ball mills, rolls, tanks, agitators, and pumps)	1,740,000	6.7
Gas handling (common feed pump and booster fans, gas ducts and ducts from plenum to absorber, exhaust gas ducts and ducts from absorber to reheater and stack)		
Waste water treatment (including ammonia-ification chambers and dist. clarifiers, effluent hold tanks, agitators, and pumps)	6,976,000	34.3
Stack gas reheat (in indirect steam reheaters)	1,282,000	4.9
SO ₂ scrubber (incinerator disposal facilities including feed tank, agitator, slurry disposal pumps, and pump water return pumps)	1,356,000	6.4
Subtotal	15,075,000	75.7
Services, utilities, and miscellaneous	1,185,000	4.5
Total process areas excluding pond construction	24,035,000	50.2
Pond construction	5,145,000	19.8
Total direct investment	29,000,000	100.0
Indirect investment		
Engineering design and supervision	1,407,000	4.8
Investment and engineering contractor	280,000	1.0
Construction margin	3,617,000	13.9
Contingency fees	1,142,000	4.4
Total indirect investment	6,446,000	23.9
Contingency	6,447,000	24.6
Total fixed investment	35,447,000	140.7
Other Capital Charges		
Allowance for start-up and start-up costs	3,394,000	12.9
Interest during construction	4,032,000	11.9
Total depreciable investment	43,873,000	179.5
Land	1,036,000	4.0
Working capital	1,021,000	3.9
Total capital investment	45,930,000	167.4

^a Basis: Midwest plant location represents project beginning mid-1977, ending mid-1980. Average cost basis for scaling, mid-1975. Stack gas reheat to 175°F by indirect steam reheat. Minimum in-process storage; only pumps are shared. Disposal pond located 1 mi from power plant. Investment requirements for flyash removal and disposal excluded; FGD process investment estimate begins with common feed plenum downstream of the ESP. Construction labor shortages with accompanying overtime pay increase not considered.

Source: (2,p.56)

TABLE 4.10

TOTAL AVERAGE ANNUAL REVENUE REQUIREMENTS-REGULATED UTILITY ECONOMICS^a

(500-Mw new coal-fired power unit, 3.55 S in coal; 1.2 lb SO₂/Mbtu heat input allowable emission; on-site solids disposal)

	Annual quantity	Unit cost, \$	Total annual cost, \$	Percent of total annual revenue requirements
Direct Costs				
Raw materials				
Limestone	150,300 tons	7.00/ton	1,052,100	7.86
Total Raw Material Cost			1,052,100	7.86
Conversion Costs				
Operating labor and supervision	25,990 man-hr	12.50/man-hr	324,800	2.30
Utilities				
Steam	489,300 Mbtu	2.00/Mbtu	978,600	6.94
Process water	247,400 kcal	0.12/kcal	29,680	0.21
Electricity	56,670,000 kWh	0.025/kWh	1,416,750	11.56
Maintenance				
Labor and material			1,822,600	12.92
Analyses	3,760 man-hr	17.00/man-hr	63,920	0.45
Total Conversion Costs			4,663,500	34.05
Total Direct Costs			5,715,600	42.15
Indirect Costs				
Capital Charges				
Depreciation, interest, replacement, and insurance at 6.0% of total depreciable investment			2,800,600	19.66
Average cost of capital and taxes at 6.0% of total capital investment			4,150,600	29.72
Overheads				
Plant, 50% of conversion costs less utilities			1,105,800	7.84
Administrative, 10% of operating labor			32,480	0.23
Total Indirect Costs			8,059,000	57.85
Total Annual Revenue Requirements			14,100,700	100.00
Equivalent unit	Mills/kWh/	8/ton coal burned	3/mbtu heat input	5 reserves \$ received
revenue requirements	4.63	9.40	0.45	46.01

^a Basis: Midwest plant location, 1950 revenue requirements. Peashing life of power plant, 30 yr. Power unit on-stream time, 7,000 hr/yr. Coal burned, 1,500,100 tons/yr, 9,000 Btu/kWh. Stack gas reheat to 175°F. S removed, 34,500 short tons/yr; solids disposal 192,000 tons/yr calcium solids including only hydrate water. Investment and revenue requirement for removal and disposal of flyash excluded. Total direct investment, \$45,000,000; total depreciable investment, \$46,877,000; and total capital investment, \$48,726,000.

Source: (2,p.56)

TABLE 4.11 Distribution of Estimated Capital Costs for FGD Systems						
Category	Process	Limestone Wet Scrubbing 500 mw (1972)	Double Alkali Scrubbing 500 mw (1974)	Magnesia Scrubbing 500 mw (1972)	Sodium(W-L) Scrubbing 500 mw (1975)	
I.	Gas handling area: fans, reheat system, scrubbers, duct work	3,401,000	3,072,000	4,483,000	7,695,000	
II.	Chemical Systems	293,000	2,808,000	2,734,000	6,840,000	
III.	Sulfuric acid plant	---	---	2,877,000	2,565,000	
IV.	Bulk materials: Steel, concrete, piping, electrical & instrumentation work, paint, optional bypass duct, etc.	2,777,000	2,521,000	3,411,000	7,600,000	
V.	Construction Costs: Labor, indirect field costs, construction fees	5,513,000	5,800,000	3,796,000	8,740,000	
VI.	Contractor's Costs: Design engineering, fees, contingency, start-up, etc.	3,800,000	3,300,000	6,361,000	4,560,000	
VII.	Interest during Construction	629,000	---	665,000	---	
VIII.	Solids Disposal	934,000	---	---	---	
IX.	Land improvements	375,000	---	350,000	---	
	Totals:	17,722,000	17,501,000	24,877,000	38,000,000	

TABLE 4.12 Distribution of Estimated Operating Costs for FGD Systems						
Cost Category	Process	Limestone Wet Scrubbing 500 mw (1972)	Double Alkali Scrubbing 500 mw (1974)	Magnesia Scrubbing 500 mw (1972)	Sodium(W-L) Scrubbing 500 mw (1975)	
	Direct Costs:					
I.	Chemicals	\$ 394,600	\$2,193,378	\$ 137,000	\$ 956,000	
II.	Operating Labor & Supervision	139,700	638,074	235,200	216,000	
III.	Utilities	542,700	1,316,027	1,181,200	1,500,000	
IV.	Maintenance	1,149,100	438,676	1,478,600	842,000	
V.	Laboratory (Analyses)	38,000	---	85,000	72,000	
VI.	Misc. Supplies	---	---	---	45,000	
	Subtotal Direct Costs:	2,264,100	4,586,155	3,117,200	3,631,000	
	Indirect Costs:					
VII.	Capital Charges	2,140,200	1,236,268	3,121,500	2,712,000	
VIII.	Overhead	342,400	---	923,800	343,000	
	Subtotal Indirect Costs:	2,482,600	1,236,268	4,045,300	3,055,000	
IX.	On Site Solids Disposal	629,600	Unknown	None	None	
	Total Annual Operating Costs	\$5,376,300	\$5,822,423	\$7,762,500	\$6,180,000 (no sulfur credit)	

Source: (3.p.56)

TABLE 4.13

Raw Material and Utility Requirements for the
Wellman-Lord Process

Basis: 500 Mw, 3.5% Sulfur Coal,
5,260 Hours Per Year at Full Load

Processing Area	Electric Power (Mw)	Steam-H.P. (MM Btu/hr)	Steam-L.P. (MM Btu/hr)	Reducing Gas (MM Btu/hr)	Sodium Carbonate (lb/hr)	Process Water (mgph)	Cooling Water (mgph)
Pretreatment and SO ₂ Absorption	9.3	77	-	-	2,120	-	-
Regeneration	1.6	-	160	-	-	25.5	430
Purge Treatment	1.0	-	20	-	-	-	-
SO ₂ Conversion	0.3	-	-(12)	92	-	-	-
TOTAL	12.2	77	166	92	2,120	25.5	430

Source: (1.p.256)

TABLE 4.14

Comparative Capital Investments
for Power Systems (\$/M)

Equipment Costs	PC Boiler with Flue Gas Desulfurization	Atmospheric FBC Boiler	Pressurized FBC Boiler
Steam gen./combustor	31.5	25.1	14.5
SO ₂ /NO _x control	30.0	3.7	14.5
Particulate removal	10.0	10.0	12.0
Coal handling & feeding	13.6	11.0	10.5
Piping, flue, & ducts	2.6	1.7	4.4
Air heater	3.3	4.6	0.0
Other boiler plant	7.6	9.7	6.0
Total Boiler Plant/Combustor	50.6	65.6	52.7
Gas Turbine-Generator	-	-	14.6
Steam Turbine-Generator	55.8	55.0	44.1
Structures & Improvements	27.3	24.8	18.5
Other Plant Consts. (electrical)	53.2	53.3	49.3
Total Equipment	234.1	155.9	177.4
Contingency	14.1	11.9	10.6
Interest & Escalation	107.2	86.4	65.0
General & Engineering	5.5	5.5	5.5
Total	352.9	302.7	260.7

NOTE - 500 Mw plant, 1976 dollars, throwaway products from FGD and FBC

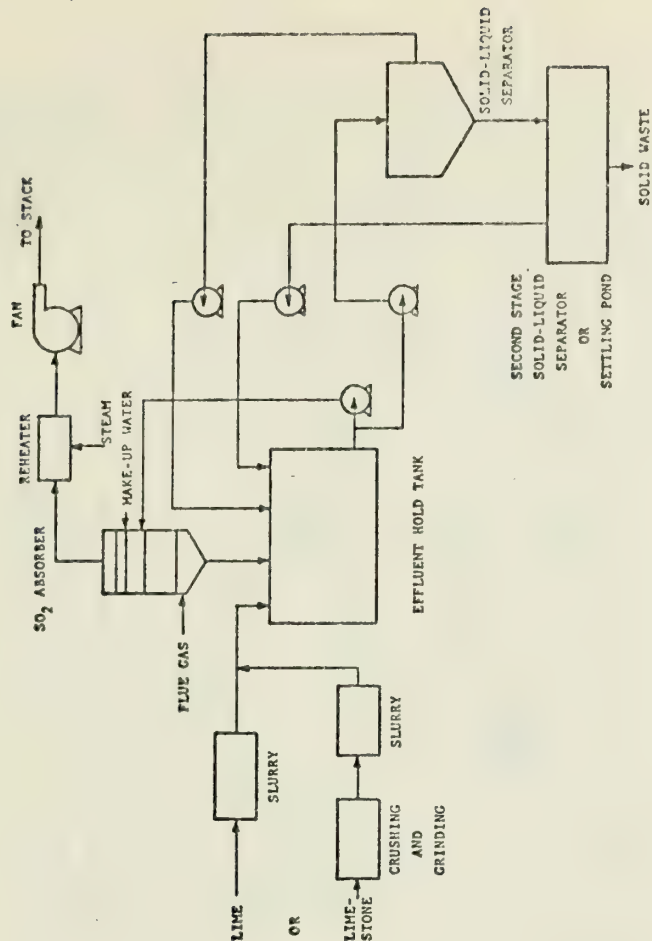
Source: (1.p.5-13)

TABLE 4.15 Comparative Operating Costs for Power Systems (\$/kW/yr)			
	PC Boiler with Flue Gas Desulfurization	Atmospheric FBC Boiler	Pressurized FBC Boiler
Capital Charges @ 15 percent	7.56	6.46	5.36
Maintenance & Operating	0.99	0.67	0.71
Fuel @ 4¢/MMBtu	3.60	3.60	3.60
Dolomite or Limestone (makeup & disposal)	0.60	1.10	1.10
TOTAL	12.75	11.82	10.77

NOTE - 7000 hrs./year

Source: (1,p.5-14)

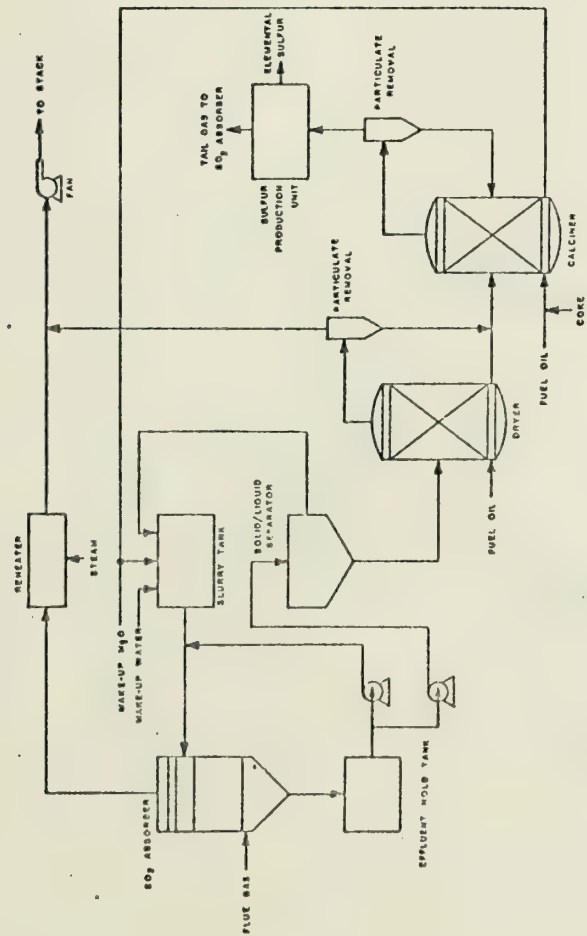
FIGURE 4.1.



Process Flow Diagram for Lime/Limestone Scrubbing Process

Source: (1,p.337)

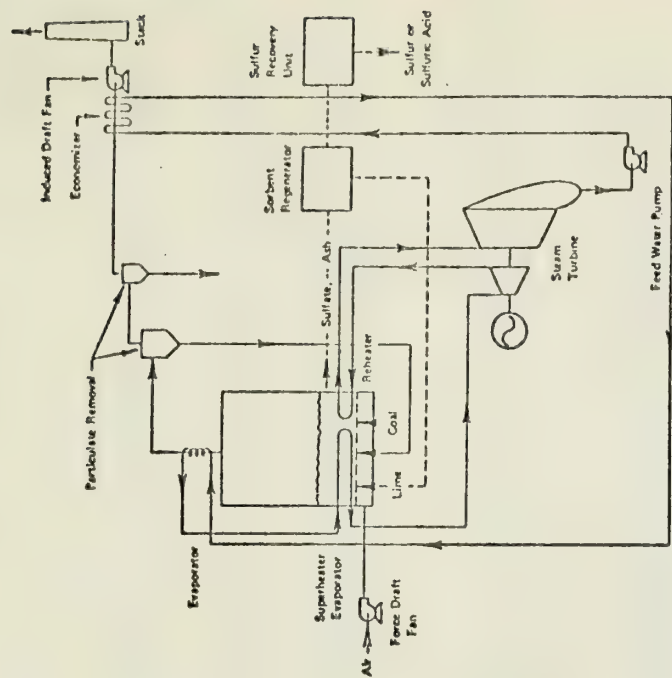
FIGURE 4.4.



Process Flow Diagram for the Magnesia Slurry Absorption Process

Source: (1,p.311)

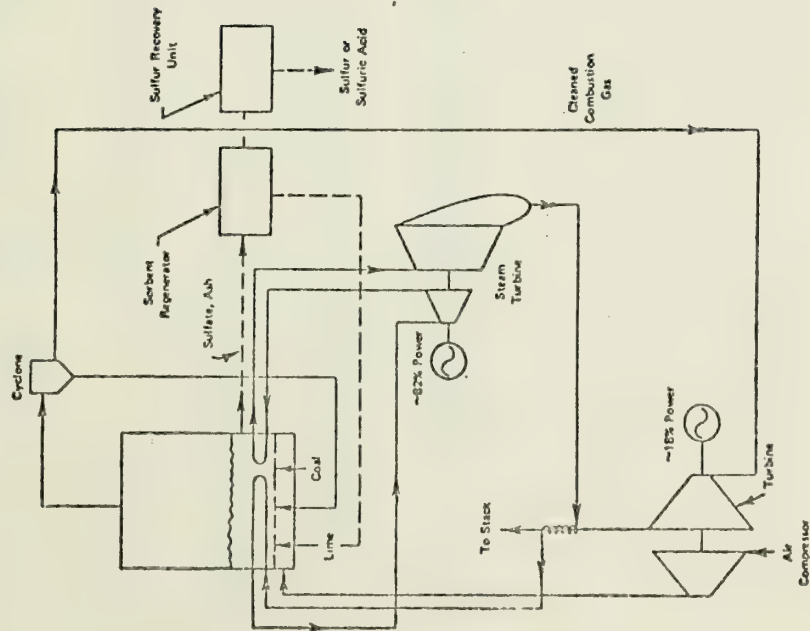
FIGURE 4.5



Atmospheric Fluidized-Bed Combustion Power Plant

Source: (1.p.5-7)

FIGURE 4.6



Pressurized Fluidized-Bed Combustion Power Plant

(1,p.5-8)

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5.0 EXTRA HIGH VOLTAGE TRANSMISSION AND DISTRIBUTION

There are three possible modes for moving coal-energy to and from the ORBES region. They are rail transport, slurry pipelines, and extra high voltage transmission. The first two methods deal with the physical movement of coal, whereas the third deals with moving the power product. In extra high voltage (EHV) transmission the coal is transformed into electricity in the mining region and sent to distant consumers. The feasibility of this idea was demonstrated during the coal strike this past winter.

There are two types of power movement available - AC and DC transmission. AC transmission has been our leading power mover since the development of the AC transformer. This allowed the transport of large voltages over greater distances. But there are characteristic line losses in this system which can be as much as 33 percent greater than for a comparable DC system [1,p.90]. Improved insulator and conductor bundle design may improve the transmission efficiency of an AC system but, at this time, it appears that for distances greater than 600 miles the DC system would be a better choice (even with the high cost of terminal equipment). This is illustrated in Figure 5.1 [1,p.95]. Some of the more desirable features of a DC system are given below [1,p.91]:

9. "Summary Evaluation of Atmospheric Pressure Fluidized Bed Combustion Applied to Electric Utility Large Steam Generators." Electric Power Research Institute. prepared by Babcock and Wilcox, October 1976.
10. "Developing Fluidized Bed Combustors for Process Heaters Under Way," The Oil and Gas Journal, January 9, 1978.
11. Fluidized Bed Boiler Completes Test Run." Department of Energy Information - Weekly Announcements, Vol. 2, No. 22, June 2 1978. p.2.
12. Cox, N.D., Richardson, A.S. and Jensen, W.P., "Energy Oriented Study of Fluidized Bed Systems " EG and G Idaho, Inc., Idaho Falls, May 1977.
13. Farmer, M.H., et al., "Application of Fluidized-Bed Technology to Industrial Boiler." Exxon Research and Engineering Co. Linden N.J. January 1977.

1. Greater power per conductor
2. Lighter and simpler line and tower construction
3. A ground return can be used-each conductor can be operated as an independent circuit
4. No charging current
5. No skin effect
6. The cables can be worked at a higher voltage gradient
7. The line does not require reactive compensation
8. There is less corona loss and interference
9. Distance is not limited by stability
10. A DC system may interconnect with AC systems of different frequencies
11. There is low short circuit current
12. Tie-line power is easily controlled

It appears that a DC transmission system would be the obvious choice for delivering bulk power loads over great distances. Tables 5.1 (1,p.98) and 5.22 (1,p 99) list some relative costs for representative AC and DC transmission systems, respectively. Note that the capital investments for a DC transmission system are substantially lower, as are the annual fixed costs. Although

the operating costs are slightly higher, the overall annual charges do indicate a rather large potential savings for this distance.

For the shorter distances-such as distribution lines within the ORBES region-AC systems would be preferable. The primary difference between transmission and distribution systems is simply their function [2,p.I-1]. Transmission systems deal with moving large blocks of power to major load centers, whereas distribution systems are responsible for voltage reduction and circuit subdivision. Thus, large demands could effectively be met by an AC grid, but the relatively lower voltages would reduce their inherent, and uneconomical, line losses.

A DC transmission system feeding into an AC distribution network would appear to be the optimum design. But the key concept here is that of decentralization of the power supply. This system not only has the ability to import to, or export from, the ORBES region large amounts of power with minimal line loss, but it can also direct this voltage in substantial amounts through numerous regional interconnects. This provides greater mobility, at the lowest cost, for supplying various amounts of power to different sectors of a region. To go a step further, future power plant reserve capacities could be looked at as an overall network supply and could be reduced.

TABLE 5.1		
COST ANALYSIS OF 765 KV ^a BW AC TRANSMISSION SYSTEM (1975 dollars) (3000 MW, 1000 Mile)		
I. CAPITAL INVESTMENTS:		
A. Sending Substation	35,651,600	
B. Receiving Substation	41,306,900	
C. Compensation		
C1. Series Capacitors @ \$14.0/kvar	59,500,000	
C2. Shunt Reactors @ \$16.0/kvar	79,050,000	
C3. Shunt Capacitors @ \$10.85/kvar	15,942,500	
D. Transmission line facilities @ \$353,600/mile	353,600,000	
E. Right of way @ \$1500/acre	40,500,000	
Total Capital Costs:	625,141,000	
II. ANNUAL FIXED COSTS:		
A. Sending Substation @ 13.9% of IA	4,955,572	
B. Receiving Substation @ 13.9% of IB	5,741,659	
C. Compensation @ 13.9% of IC	21,408,961	
D. Transmission facilities @ 13.5% of ID	47,745,167	
Total Annual Fixed Costs:	79,850,339	
III. ANNUAL OPERATING COSTS:		
A. Sending Substation @ 2.3% of IA	834,247	
B. Receiving Substation @ 2.6% of IB	1,073,919	
C. Compensation @ 1.3% of IC	2,001,902	
D. Transmission facilities @ 1.3% of ID	4,597,868	
E. Electrical Energy Losses:		
E1. Energy Losses @ 12/kwh	37,920	
E2. Power Losses @ \$180/kwh & 13.9% fixed chrgs		
Total Annual Operating Costs:	8,546,024	
IV. ANNUAL COSTS: (II + III)		
V. UNIT COSTS:		
A. Mills/ton-mile of equivalent coal		7.641
B. Mills/kwh-r received power		3.843
C. Investment \$/kwh-r received power		27.2

^aDouble circuit is used.

^bBased on EPC P-3b Annual fixed charge rate.

^{**}Includes administrative and general expenses @ 3.0% of their operating costs.

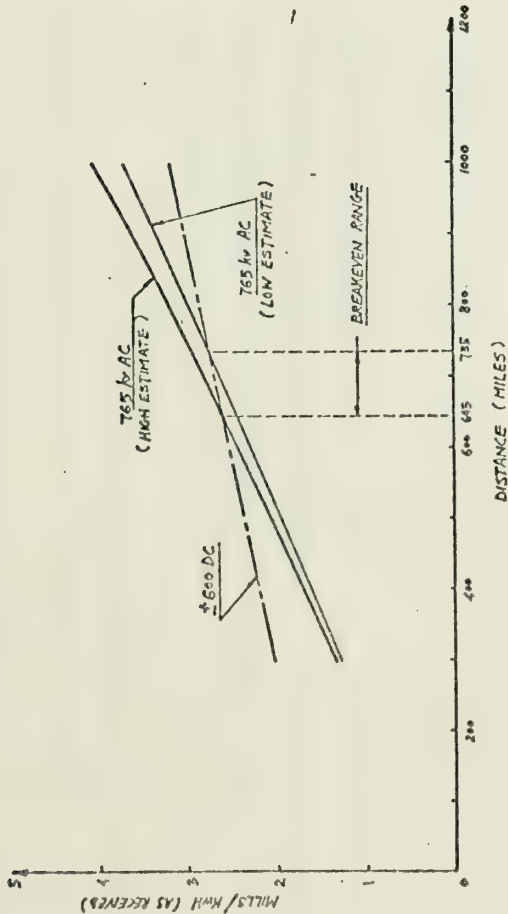
Source: (1.p.98)

TABLE 5.2		
COST ANALYSIS OF 765 KV BW-DC TRANSMISSION SYSTEM (1975 dollars) (3000 MW, 1000 Mile)		
I. CAPITAL INVESTMENTS:		
A. Sending Substation	123,000,000	
B. Receiving Substation	118,712,000	
C. Transmission line facilities @ \$236,000/mile	236,000,000	
D. Right of way @ \$1500/acre	31,500,000	
Total Capital Costs:	509,212,000	
II. ANNUAL FIXED COSTS:		
A. Sending Substation @ 13.9% of IA	17,097,000	
B. Receiving Substation @ 13.9% of IB	16,501,900	
C. Transmission facilities @ 13.5% of ID	31,860,000	
Total Annual Fixed Costs:	65,458,900	
III. ANNUAL OPERATING COSTS:		
A. Sending Substation @ 2.3% of IA	2,876,200	
B. Receiving Substation @ 2.6% of IB	3,006,500	
C. Transmission facilities @ 1.3% of ID	3,008,000	
D. Electrical Energy Losses:		
D1. Energy Losses @ 12/kwh	30,660	
D2. Power Losses @ \$200/kwh & 13.9% fixed chrgs		
Total Annual Operating Costs:	9,003,300	
IV. ANNUAL COSTS: (II + III)		
V. UNIT COSTS:		
A. Mills/ton-mile of equivalent coal		6.255
B. Mills/kwh-r received power		3.153
C. Investment \$/kwh-r received power		21.5

Source: (1.p.99)

FIGURE 5.1

ESTIMATED UNIT COST OF EHV TRANSMISSION OF 3000 MWe (1975\$)



Estimated Unit Cost of EHV Transmission of 3000 MWe (1975\$)

Source: (1,p.95)

REFERENCES

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2. "Electric Transmission and Distribution," report for Task A of Contract OIA-C-180, submitted by Teknekron, Inc., January 18, 1978.

6.0 RECOMMENDATIONS -- I-O ANALYSIS

The methodology applied here is that of Herendeen (1) with handbook examples provided by Bullard (2). This involves constructing column matrices of technical coefficients showing the relative operating costs of future coal technologies. As these are energy coefficients, they are expressed in Btu/Btu for energy sectors and \$/Btu for non-energy sectors.

The first step is the categorization within the Ohio River Basin Energy Study region (ORBES) of the costs involved in the various technologies. The ORBES demand study has established a 48 by 48 coefficient matrix which separates and aggregates various aspects of the ORBES region economy. The SIC classifications for the ORBES IO model are shown in Table 6.1. From this list it is determined which classifications apply to the energy technologies considered here.

The next step is to examine the subdivisions of the applicable SIC classifications to determine the cost considerations more specifically. Up to this point all eliminations have been due simply to their inapplicability to any of the processes considered. That is, nothing has been discounted due to specific selection of costs requiring documentation. Note also that in Table 6.1 the ORBES

classifications are listed such that the SIC classifications are divided into three groups listed in numerical order. Each group exhibits a decreasing energy sensitivity in relation to the ORBES economy. Thus no SIC classifications were subdivided past ORBES group number 30; costs past this point, where applicable, are relatively constant. In addition, the second sequence contains many stable costs. This is especially true of groupings numbered 5000 and above (wholesale and retail trade, finance, insurance, real estate and services). Therefore, the major differences among the various technologies occurs in ORBES groups 1-5, which include SIC classifications 1100 to 5000 (coal mining, construction, manufacturing, transportation, communication, and electricity, gas, and sanitary services).

The evaluation procedure to be followed for the relatively insensitive categories is analogous, where more specific disaggregation is utilized in distinguishing the new technologies. For example, consider furniture requirements for the offices associated with plants. Because commercial plants, e.g. for gasification or liquefaction, are still in the future, specific furniture requirements can only be approximated. For this insensitive category, sufficient approximation can be obtained by utilizing the office furniture requirements for a similar type of plant operation. This procedure can also be applied to categories such as heating, some construction,

transportation, etc.

Alternatively, consider low-Btu gasification processes utilizing a fluidized-bed, air-blown, ash-agglomerating process as the most feasible future technique. Currently the major drawback with these processes are operating characteristics involving coal feed and fines handling which can affect the efficient use of the coal. Thus specific consideration must be given to related materials and labor requirements. However, the use of low-Btu processes will be in large industrial parks for process heat, or as a boiler fuel in steam power plants. Thus many of the costs are analogous to contemporary steam generator operation and supervision. In addition, these costs may be extrapolated to high-Btu gasification operations (also utilizing a fluidized bed) to include oxygen requirements, shift conversion, and methanation steps. The most common plant size studied is one producing 250 MMscfd of gas.

A similar analogy can be drawn for liquefaction and other processes. In the case of liquefaction a chemical processing plant can be used, or a refinery producing 50,000 bpd. The use of a refinery is advantageous in that a potential design application of liquefaction is the "coalfinery" described previously. Fluidized bed combustion also utilizes steam generation facility data, with coal handling and dolomite supply as additional operational costs. Combined cycle configurations

simply utilize gasification cost results and incorporate them with given data on steam and gas turbine power plants. The same may be said for general applications of cogeneration units. For lime/limestone FGD systems, some adjustments, such as M_{gO} supply, have to be made for the magnesia slurry technique. Extra high voltage transmission costs can be approximated from existing data, as essentially conventional technology used for AC or DC systems.

Upon selecting the technologies with potential commerciality, the problem becomes one of determining the costs incurred in these processes. This involves an economic and engineering analysis of the necessary equipment. More importantly, it involves discriminatory utilization of applicable verified information. New ideas are still founded on what is currently available. The next step is separation by process of the general costs (for example, coal mining would not be a cost incurred by an FGD system). From this energy costs may be determined in the appropriate manner for sensitive and insensitive categories. This analysis is facilitated by a Brookhaven report by Lukachinski and Tessmer (3). This report contains technical coefficients for energy technologies projected to 1985. This is appropriate for our purposes as 1985 may be taken as a key transition point for the processes proposed. The non-energy coefficients are normalized in terms of 1967 dollars. The basic IO energy study utilized a 110-sector economy as shown

in Table 6.2 (3,p.3). However, this in turn was reduced from a 367-sector IO energy matrix based on BEA data. The 48-sector ORBES energy demand study is consistent with the above. This makes aggregation a simple case of bookkeeping. Much of the information given in the report is based on the aforementioned MITRE study (4), which is also subdivided according to the 367-sector BEA economy. In some cases this study are referred to directly. This chapter only discusses sectors undergoing change. Other sectors' documentation will simply be referenced, as they are included in the overall ORBES matrix.

As an example, consider high-Btu coal gasification. The method used in the Brookhaven report was the Hygas process, which is the representative process chosen in this report for the future. The coefficients given in this report (3,p.13) remain essentially the same for our purposes, especially the non-energy requirements. One possible change could be that of the chemical feedstock but, as shown in Table 2.9, this makes up less than 3% of the cost and may be taken as given. The change that must be made is in the type of coal used. We are interested in the bituminous coal requirements rather than those of lignite. This can be demonstrated by proportionately adjusting the Btu requirement of input coal in terms of relative Btu ratings of equal tonnage of bituminous and lignite coal. Comparing the dry heating values of Illinois bituminous and lignite coal (5,p.13), it is seen that the coal requirement should be reduced by 16%.

It is conceded that lignite is more reactive and cleaner, but it must be assumed that operational problems are resolved once the process is ready for commercial application.

The representative low-Btu gasification process being considered is U-Gas. Both Hygas and U-Gas are techniques developed by IGT and have many similar operational and design characteristics, resulting in a good deal of similarity in the technical coefficients. The Hygas coefficients are used as a baseline for this analysis. For example, coal handling and pretreatment costs are similar for both processes, as well as purification costs. At first one might think coal requirements would be drastically different. This, however, is not the case as both Table 1.3 and the MITRE report (4,p.220) show similar operational efficiencies. Naturally U-Gas requires more coal recycling to obtain as much Btu output, but this again is part of the resolvable operational considerations. The important differences in coefficients occur due to hydrogen production and methanation requirements. Because CO conversion and methanation catalysts are removed for the low-Btu gasification process, they may be eliminated from the chemical requirements (4,p.210) and hence reduce section 50 of the Brookhaven model by 54%. Note that the adjustment has again been made proportionately, as we are dealing with ratios. The hydrogen-producing technique used in the given study was the electrothermal method. Although this may not be the technique used in the future, elimination of this

can give a representative reduction in cost. Elimination of the Btu requirement for power generation (assuming the majority is required for the electrothermal process) results in a coal coefficient (Sector 1) reduction of 6%. Note that this is applied to the coal sector, as coal is used for the Btu input. Recall that this modification is applied to the changed Hygas coefficient mentioned above.

Brookhaven also conducted a study of the SRC liquefaction process (3,p.20), utilizing much of the MITRE data. Here the SRC product has been considered as a boiler fuel product, which is where much of the emphasis is being placed for the proposed demonstration plants. As mentioned before, the product may later be used as a refinery feedstock, which may require some of the coefficients to be altered and/or incorporated into a refinery operation. The non-energy coefficients given are representative for our purposes at this time. In addition, because a Pittsburgh seam coal is used for this study, and refinery and coal gasification coefficients are used where applicable, the energy coefficients are also useful for our purposes.

But this only demonstrates the requirements for a solvent extraction technique (material is not yet available, but it may be assumed that the coefficients for the Exxon Donor-Solvent technique are similar to those of SRC). For a hydrogenation technique, such as H-coal, a few changes in the coefficients are

required. As the principle applied is generally the same for both techniques, we will assume similar non-energy coefficients. The major differences are due to larger coal (Sector 1) and chemical (Sector 50) requirements. This is demonstrated by the requirements given in Tables 2.20 and 2.23. Assuming the same cost and characteristics for the materials required, these two sectors may be adjusted proportionately. The coefficient for coal increases 15% and the sector for chemicals by 159% over those of SRC. All other coefficients are considered applicable as given.

The technical coefficients for a limestone slurry process, the most extensively used FGD system, are taken from Table 4.10. The lime slurry FGD system is also used, but its requirements are essentially the same as those for limestone. More operational information is required to do a study of the regenerative magnesia slurry technique, but Table 4.11 illustrates some of the relative cost considerations. Note the savings for solids disposal.

It must be pointed out that the FGD system is not an entity by itself. Its coefficients are incorporated into the operation of a plant where cleanup is required. The coefficients for a limestone process, derived from data in Table 4.10 in terms of the 110-sector model, are given below:

(16)	.0409	Btu/Btu
(20)	.0162	Btu/Btu
(35)	.2803	\$/MMBtu
(50)	.0927	\$/MMBtu
(100)	.5851	\$/MMBtu

The technical coefficients presented by the Brookhaven model for combined cycle operation seem to be the best available. These are incorporated from NITRE data as explained in their report. This involves the incorporation of a COGAS topping system with an entrained bed low-Btu gasification system. This topping system is representative of possible future application. In addition, a hot-gas cleanup system is utilized, which must be assumed economically feasible as high temperature gases are required if substantial output is to be available. Note that we assume resolution of the high-temperature turbine blade problems. A fluidized-bed gasifier may be preferred to the entrained-bed variety, but many of the advantages are present with this system and are reflected in the coefficients. For comparison a coefficient from Sector 1 was determined for the EAH case of Table 3.8 for total potential output (i.e., 1000 Mw) and was found to be 2.108.

In the Brookhaven report (3,p.16), the technical coefficients for an FEC system are essentially the same as those for a coal-fired electric generation plant. Note that these

coefficients are for utility boiler use. The major difference here occurs for maintenance and operation (Sector 35) and chemical requirements (Sector 50). These adjustments are made from relative costs given in Table 4.15. This table, taken from a Teknekron paper (6.p.5-14), compares a pulverized bed combustion boiler to atmospheric and pressurized fluidized bed systems. Fuel costs have been normalized to equity in that table; that is, efficiency advantages are incorporated with maintenance and operating considerations. Thus there is no change in Sector 1; we assume the same Btu values are extracted in each case at the expense of increased operating costs. It must be remembered that the values given in this table are relative values, but the initial coefficients from the Brookhaven report are viewed as reasonably accurate. From Table 4.15 it is seen that Sector 35 will decrease by 32% for the atmospheric case and 28% for the pressurized case. Sector 50 will increase by 83% for both cases. Note that it is assumed here that the greatest effect on changing technologies is operational procedure. Note also that an additional change will occur under finance and insurance charges (Sector 100). This will decrease by 14% for the atmospheric case and 29% for the pressurized case.

EHV transmission may be considered here as a cost of utilization or a mode of potentially competitive imports or exports. That is, aside from fixed charge rates on the equipment, and manual operational expenses, there is no

technological breakdown necessary for annual operating costs. This is simply a process for delivering generated power from one location to another, and is entered in toto under Sector 77 of the Brookhaven model. We use as a standard example (the values given by Soo and Rieber (7,p.99) in Table 5.2) for a #60KV EHV-DC transmission system delivering 3000 MW over 1000 miles. The values in this table result in a technical coefficient with a value of 1.03944 dollar per million Btus.

The values used here are the result of rather extensive and detailed studies of applicable coal technologies. The changes made not only alter these coefficients for our specific purposes, but also illustrate some of the more important considerations related to the operating costs of these processes. Thus we not only selected a generic process with feasible future applications from among the prominent technologies available, but have been able to give a comparable prediction of costs for these technologies. Recall again that we assumed resolution of various equipment and operational problems in some cases. For example, we assumed development of an effective hot-gas cleanup system and a high-temperature turbine blade for a combined cycle power plant. In addition we assumed development of a pressurized fluidized bed reactor, with favorable results compared to the atmospheric case. The main argument against the processes in this study is the actual time horizon we are considering. This also applies to the future regenerable FGD systems as much of the

future plans involve contemporary "throw-away" techniques. Hopefully this trend may change and make direct combustion systems attractive and economically feasible. This is especially critical in the high-demand, high-sulfur coal, ORBES region.

But this very demand may dictate development and advocacy of coal conversion technologies. High-Btu gasification, the original conversion technology on which extensive research and development had been undertaken, appears to be the most expensive for what is received. It can only be justified on an extensive scale by a desperate demand for natural gas alone. Low-Btu gasification is much more economical and could be of great benefit in industrial as well as utility applications. A major possibility is the incorporation of a process with an industrial park. Note that the most favorable processes for gasification are of the fluidized-bed variety, as these are able to handle the predominately caking coals of the ORBES region.

Another possibility for a gasification system is its incorporation into a coal liquefaction complex. For example, it could extract additional coal products from the residual char, as was the original purpose of the COGAS technique. The most desirable future coal conversion technology for the ORBES region appears to be coal liquefaction; specifically the SRC (solvent extraction) technique. This seems to be the most economical process per unit of output. It has a high conversion efficiency,

and can handle the high-sulfur, caking coals of the ORBES region. It is also a proven process soon to be demonstrated on a large scale. Currently these plants will be designed to produce a clean boiler fuel, but future expansion should include the aforementioned multiproduct "coalfinery" concept. This would allow the ORBES region to deal not only with increases in energy demand, but variations in the fuels required in the future as well.

It is rather obvious that the technologies, in various stages of development, are present to assist in meeting the energy demands of the ORBES region. It is also obvious that development and the use of the extensive coal resources in the ORBES region can play an important part in the national energy economy. But it must be kept in mind that the type of energy demanded (e.g., for residential, industrial or utility purposes) will help dictate not only the process but the fuel required. This makes the coal liquefaction concept so attractive.

TABLE 6.1				
SIC CLASSIFICATIONS FOR 48-SECTOR ORBES MODELS				
ORBES REGION	SIC CLASSIFICATIONS			
1	1100	1200		
2	1310	1320		
3	2910	2990		
4	4910	4931		
5	4920	4932		
6	01--			
7	1000	1400		
8	15--	1350		
9	2010			
10	2000	-2010	2100	
11	2200			
12	2300			
13	2400	2500		
14	2500	-2620	-2630	-2650 2700
15	2820			
16	2830			
17	2850			
18	2810			
19	2800	-2610	-2620	
20	2820			
21	2950			
22	3000	3100		
23	3200			
24	3310			
25	3320	3390		
26	3330			
27	3340	3350	3360	
28	3400			
29	3500			

TABLE 6.1
(Continued)

SIC CLASSIFICATIONS FOR
46-SECTOR ORES MODELS

ORBS REGION	SIC CLASSIFICATIONS			
30	3600			
31	3710			
32	3700	-3710		
33	3800	3900		
34	4740			
35	4100	4600	4700	-4740 4800
	4940	4950	4960	4970
36	4200			
37	4400			
38	4500			
39	5000			
40	52--	7396		
41	6000	6100	6200	6400 6700
42	6500	6600		
43	7000	8100	8900	-8920
	7800	7900		
44	7200	7600	7300	-7310 -7396
45	7310			
46	7500			
47	8000	8200		
48	8400	8600	8920	

TABLE 6.2

SECTORAL CLASSIFICATION OF ENERGY INPUT-OUTPUT MODEL
AND ALIGNMENT WITH OTHER CLASSIFICATION SYSTEMS

BNL 110	Sector	BNL 101	BEA
1	Coal	1	7
2	Crude Oil & Gas	2	6
3	Shale Oil	-	-
4	Methane from Coal	3	-
5	Solvent Refined Coal	-	-
6	Refined Oil Products	4	31.01
7	Pipeline Gas	5	63.62
8	Coal Combined Cycle Electric	-	-
9	Other Fossil Electric	6	66.01
10	Low Electric	7	66.01
11	High Electric	-	-
12	Hydroelectric	8	66.01
13	Gre Reductions Feedstocks	9	-
14	Chemical Feedstocks	10	-
15	Motive Power	11	-
16	Process Heat	12	-
17	Water Heat	13	-
18	Space Heat	14	-
19	Air Conditioning	15	-
20	Electric Power	16	-
21	Livestock and Livestock Products	17	1
22	Other Agricultural Products	18	2
23	Forestry and Fishery Products	19	3
24	Agricultural, Forestry and Fishery Services	20	3
25	Iron and Ferroalloys Ores Mining	21	5
26	Nonferrous Metal Ores Mining	22	6
27	Stone and Clay Mining and Quarrying	23	9
28	Chemicals and Fertilizer Mineral Mining	24	10
29	New Construction, Residential buildings	-	11.01
30	New Construction, Nonresidential buildings	-	11.02
31	New Construction, Public Utilities	25	11.03
32	New Construction, Highways	-	11.04
33	New Construction, All Other	-	11.05

Code BNL 110 - 110 sector version
BNL 101 - 101 sector version
BEA - Bureau of Economic Analysis, U.S. Dept. of Commerce
Source: (3,p.3)

TABLE 6.2 (Continued)			
SECTORAL CLASSIFICATION OF ENERGY INPUT-OUTPUT MODEL AND ALIGNMENT WITH OTHER CLASSIFICATION SYSTEMS			
BNL 110	Sector	BNL 101	BEA
34	Maintenance and Repair Construction, Residential	--	12.01
35	Maintenance and Repair Construction, All Other	26	--
36	Ordnance and Accessories	--	12.02
37	Food and Kindred Products	27	11
38	Tobacco Manufactures	28	14
39	Broad and Narrow Fabrics, Yarn and Thread Mills	29	15
40	Misc. textile goods and floor coverings	30	16
41	Apparel	31	17
42	Misc. fabricated textile products	22	16
43	Lumber and wood products, except containers	33	19
44	Wooden Containers	34	20
45	Household Furniture	35	22
46	Other Furniture and Fixtures	36	22
47	Paper and Allied Products Except Containers and Boxes	37	23
48	Printing and Publishing	38	24
49	Chemicals and Selected Chemical Products	40	26
50	Plastics and Synthetic Materials	41	27
51	Drugs, Cleaning and Toilet Preparations	42	28
52	Fabrics and Allied Products	43	29
53	Farming Mixtures and Blocks	44	30
54	Asphalt Felts and Coatings	45	31.02
55	Rubber and Miscellaneous Plastics Products	46	31.03
56	Leather Tanning and Industrial Leather Products	47	32
57	Footwear and Other Leather Products	48	33
58	Glass and Glass Products	49	34
59	Stone and Clay Products	50	35

Code BNL 110 - 110 sector version
BNL 101 - 101 sector version
BEA - Bureau of Economic Analysis, U.S. Dept. of Commerce
Source: (3,p.3)

TABLE 6.2 (Continued)			
SECTORAL CLASSIFICATION OF ENERGY INPUT-OUTPUT MODEL AND ALIGNMENT WITH OTHER CLASSIFICATION SYSTEMS			
BNL 110	Sector	BNL 101	BEA
161	Primary Iron and Steel Manufacturing	52	37
162	Primary Nonferrous Metals Manufacturing	52	38
163	Metal Containers	54	39
164	Heating, Plumbing and Fabricated Structural Metal Products	55	40
165	Screw Machine Products, Bolts, Nuts, etc. & Metal Stampings	56	41
166	Other Fabricated Metal Products	57	42
167	Engines and Turbines	58	43
168	Farm Machinery	59	44
169	Construction, Mining, Oil Field Machinery, Equipment	60	45
170	Materials Handling Machinery and Equipment	61	46
171	Metalworking Machinery and Equipment	62	47
172	Special Industry Machinery and Equipment	63	48
173	General Industrial Machinery and Equipment	64	49
174	Machine Shop Products	65	50
175	Office, Computing and Accounting Machine	66	51
176	Service Industry Machines	67	52
177	Electric Trains, & Dist. Eq. & Electric Industry Apparatus	68	53
178	Household Appliances	69	54
179	Electric Lighting and Wiring Equipment	70	55
180	Radio, Television and Communications Equipment	71	56
181	Electronic Components and Accessories	72	57
182	Miscellaneous Electric Machinery, Equipment & Supplies	73	58
183	Motor Vehicles and Equipment	74	59
184	Aircraft and Parts	75	60
185	Other Transportation Equipment	76	61

Code BNL 110 - 110 sector version
BNL 101 - 101 sector version
BEA - Bureau of Economic Analysis, U.S. Dept. of Commerce
Source: (3,p.3)

TABLE 6-2 (Continued)			
SECTORAL CLASSIFICATION OF ENERGY INPUT-OUTPUT MODEL AND ALIGNMENT WITH OTHER CLASSIFICATION SYSTEMS			
BNL	Sector	BNL	BEA
110		101	62
66	Professional, Scientific & Controlling Instruments & Supplies	77	62
67	Optical, Optical, & Photographic Equipment and Supplies	76	63
68	Miscellaneous Manufacturing	79	64
69	Package and Related Services	80	65.01
90	Local, Suburban & Interurban Highway Passenger Transportation	81	65.02
91	Motor Freight Transportation and Warehousing	82	65.03
92	Water Transportation	83	65.04
93	Air Transportation	84	65.05
94	Pipeline Transportation	85	65.06
95	Transportation Services	86	65.07
96	Communications except Radio & Television Broadcasting	87	67
97	Radio and TV Broadcasting	88	67
98	Water and Sanitary Services	89	66.01
100	Finance and Insurance	91	70
101	Real Estate and Rental	92	71
102	Hotels & Lodging; Personnel and Repair Service, except Auto Repair	93	72
103	Business Services	94	73
104	Automobile Repair and Service	95	75
105	Automobiles	96	76
106	Medical, Educational Services & Nonprofit Inst.	97	77
107	Federal Government Enterprises	98	78
108	State and Local Government Enterprises	99	79
109	Business Travel, Entertainment and Gifts	100	81
110	Office Supplies	101	82

Code BNL 110 - 110 sector version
BNL 101 - 101 sector version
BEA - Bureau of Economic Analysis, U.S. Dept. of Com-
merce
Source: (3.P.3)

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